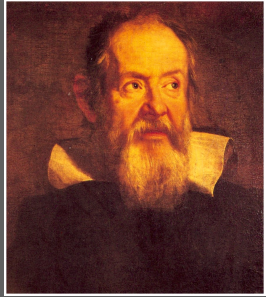


1

Determining the Speed of Light

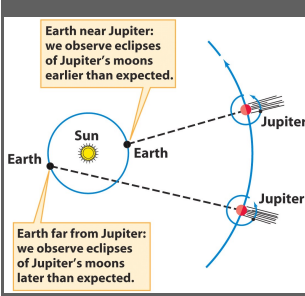
- Galileo tried unsuccessfully to determine the speed of light using an assistant with a lantern on a distant hilltop



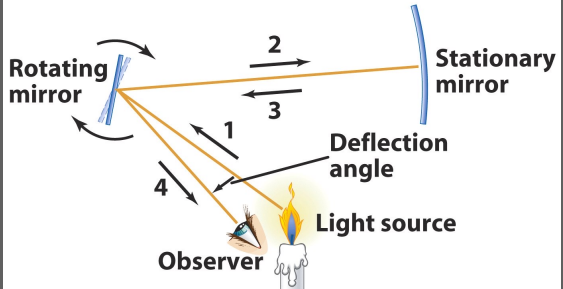
2

Light travels through empty space at a speed of 300,000 km/s

- In 1676, Danish astronomer Olaus Rømer discovered that the exact time of eclipses of Jupiter's moons depended on the distance of Jupiter to Earth
- This happens because it takes varying times for light to travel the varying distance between Earth and Jupiter
- Using $v=d/t$ with a known distance, d , and a measured time, t , gave the speed, v , of the light



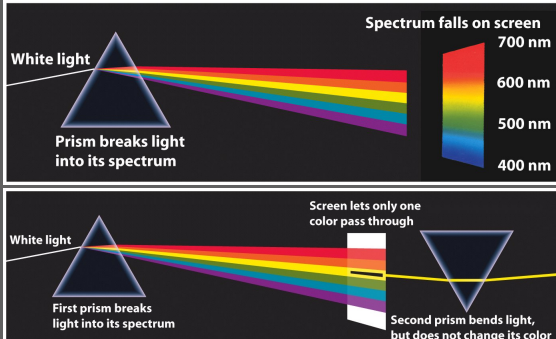
3



- In 1850 Fizeau and Foucault also experimented with light by bouncing it off a rotating mirror and measuring time
- The light returned to its source at a slightly different position because the mirror has moved during the time light was traveling
- $\Rightarrow c$

4

Light is electromagnetic radiation and is characterized by its wavelength (λ)



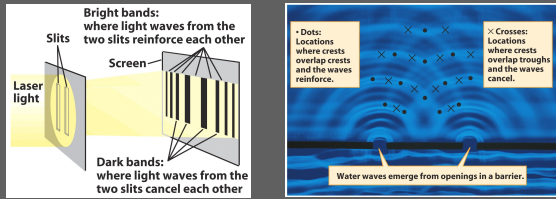
Spectrum falls on screen
700 nm
600 nm
500 nm
400 nm

White light
Prism breaks light into its spectrum

White light
First prism breaks light into its spectrum
Screen lets only one color pass through
Second prism bends light, but does not change its color

5

Light has properties of both waves and particles



- Newton thought light was in the form of little packets of energy called photons and subsequent experiments with blackbody radiation indicate it has particle-like properties
- Young's Double-Slit Experiment indicated light behaved as a wave
- Light has a dual personality; it behaves as a stream of particle like photons, but each photon has wavelike properties

6

The Nature of Light

Magnetic field Electric field

Direction of propagation

Wavelength λ

- In the 1860s, the Scottish mathematician and physicist James Clerk Maxwell succeeded in describing all the basic properties of electricity and magnetism in four equations
- This mathematical achievement demonstrated that electric and magnetic forces are really two aspects of the same phenomenon, which we now call **electromagnetism**

http://www-groups.dcs.st-and.ac.uk/~history/PictDisplay/Maxwell.html

7

Wavelength and Frequency

White light

First prism breaks light into its spectrum

Screen lets only one color pass through

Second prism bends light, but does not change its color

Frequency and wavelength of an electromagnetic wave

$$v = \frac{c}{\lambda}$$

v = frequency of an electromagnetic wave (in Hz)

c = speed of light = 3×10^8 m/s

λ = wavelength of the wave (in meters)

8

Shortest wavelength

Longest wavelength

- Photon energy
- $E = hc/\lambda = h\nu$
- $h = 6.67 \cdot 10^{-34}$ Js (Planck's constant)
- Because of
 - Visible light falls in the 400 to 700 nm range
- Stars, galaxies and other objects emit light in all wavelengths

9

Optics and Telescopes

10

Telescopes

- The fundamental purpose of any telescope is to **gather more light** than the naked eye can
- In many cases telescopes are used to produce images far **brighter and sharper** than the eye alone could ever record

11

A refracting telescope uses a lens to concentrate incoming light at a focus

Distant light source

Essentially parallel light rays

Lens

Focal point

12

How Light Beams Behave

(a) How cars behave

(b) How light beams behave

- As a beam of light passes from one transparent medium into another—say, from air into glass, or from glass back into air—the direction of the light can change
- This phenomenon, called **refraction**, is caused by the **change in the speed of light**

13

(a) **(b)**

14

15

Powers of telescopes

- Magnification
- Light gathering power
- Resolving power

17

Refracting Telescope and Magnification

Light from a distant object

Objective lens

Focal plane of objective

Focal length of objective

Objective forms an image in the focal plane

Focal length of eyepiece

Eyepiece lens magnifies the image for easier viewing

Figure 15-19
Discovering the Universe, Eighth Edition
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Magnification, m , is given by the ratio between the focal length of objective, f_0 , and the eyepiece focal length, f_e .

$$m = f_0 / f_e$$

Example: $f_0 = 4 \text{ m}$, $f_e = 1 \text{ cm} \Rightarrow m = 400$

18

Light Gathering Power

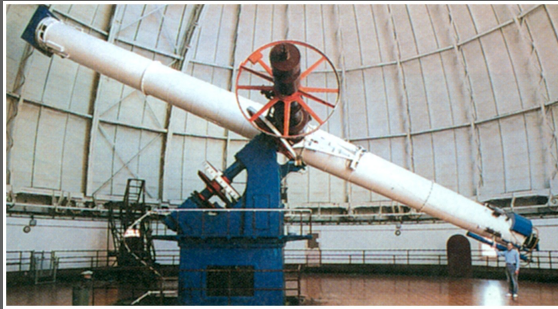
Small-diameter objective lens:
dimmer image, less detail

Large-diameter objective lens:
brighter image, more detail

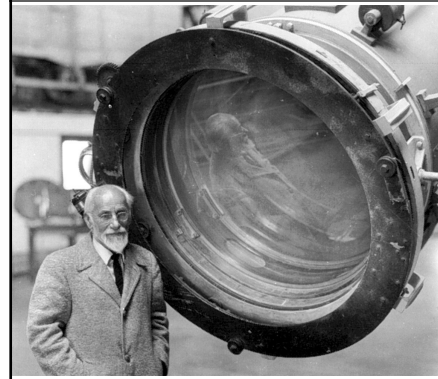
The **light-gathering power** of a telescope is directly proportional to the area of the objective lens, which in turn is proportional to the square of the lens diameter

20

Yerkes Observatory Refractor



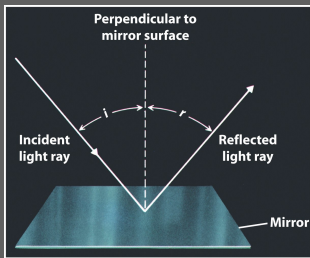
22



Glass impurities, chromatic aberration, opacity to certain wavelengths, and structural difficulties make it inadvisable to build extremely large refractors

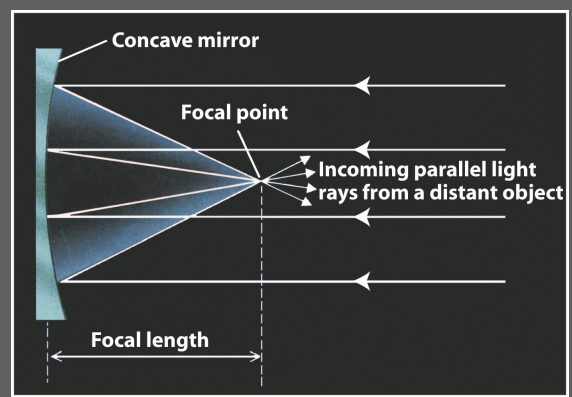
23

A reflecting telescope uses a mirror to concentrate incoming light at a focus



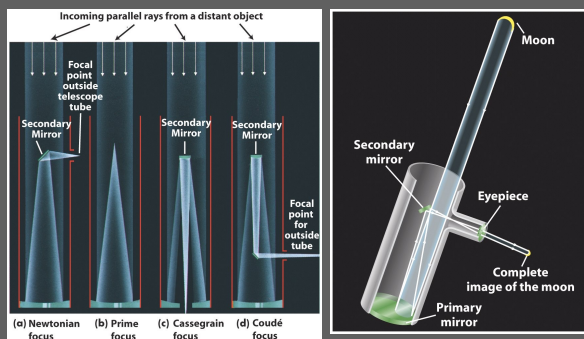
- Reflecting telescopes, or reflectors, produce images by reflecting light rays to a focus point from curved mirrors.
- Reflectors are not subject to most of the problems that limit the useful size of refractors.

24



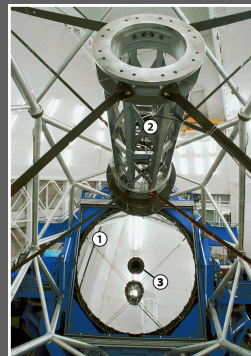
25

Reflecting Telescopes



26

Gemini North Telescope



1. The 8.1-meter objective mirror
2. The 1.0-meter secondary mirror
3. The objective mirror

27

Table 6-1 The World's Largest Optical Telescopes			
Telescope	Location	Year of completion	Mirror diameter (m)
Gran Telescopio Canarias	La Palma, Canary Islands, Spain	2004	10.4
Keck II	Mauna Kea, Hawaii	1996	10.0
Keck I	Mauna Kea, Hawaii	1993	10.0
Hobby-Eberly Telescope	McDonald Observatory, Texas	1998	11.0*
South African Large Telescope	Sutherland, South Africa	2004	9.2
Large Binocular Telescope	Mount Graham, Arizona	2004-05	Two 8.4
Subaru	Mauna Kea, Hawaii	1999	8.3
VLT UT 1-Antu	Cerro Paranal, Chile	1998	8.2
VLT UT 2-Kueyen	Cerro Paranal, Chile	1999	8.2
VLT UT 3-Melipal	Cerro Paranal, Chile	2000	8.2
VLT UT 4-Yepun	Cerro Paranal, Chile	2000	8.2
Gemini North (Gillett)	Mauna Kea, Hawaii	1999	8.1
Gemini South	Cerro Pachón, Chile	2000	8.1

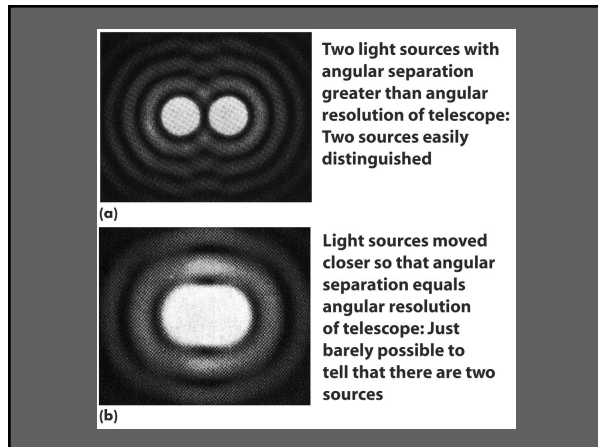
*The objective mirror of the Hobby-Eberly Telescope is 11.0 m in diameter, but in operation only an area of 9.2 m in diameter is used to collect light.

29

Telescope images are degraded by the blurring effects of the atmosphere and by light pollution

- **Angular Resolution:** A telescope's angular resolution, which indicates ability to see fine details, is limited by many factors.
- Diffraction is an intrinsic property of light waves.
- Its effects can be minimized by using a larger objective lens or mirror and/or a smaller wavelength of observed light.

30



31

Diffraction limited angular resolution

$$\Theta = 2.5 \times 10^5 \lambda / D$$

where Θ is the angular resolution in seconds of arc
 λ is the wavelength of light in metres
 D is the diameter (of mirror or lens) in metres

32

What is the diffraction-limited angular resolution of our 1-m telescope?



Planewave Instruments

33

What is the diffraction-limited angular resolution of our 1-m telescope?



Planewave Instruments

$$\begin{aligned} \Theta &= 2.5 \times 10^5 \lambda / D \\ &= 2.5 \times 10^5 \cdot 500 \times 10^{-9} / 1 \\ &= 0.125 \text{ arcsec} \end{aligned}$$

34

Powers of telescopes

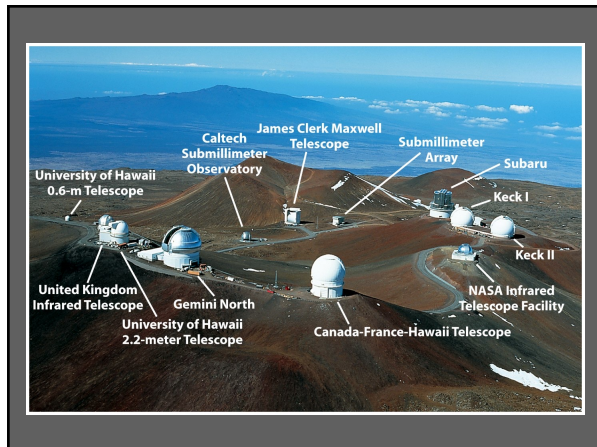
- Magnification $m = f_o/f_e$
- Light gathering power $LGP \propto D^2$
- Resolving power $\Theta = 2.5 \times 10^5 \lambda / D$

35

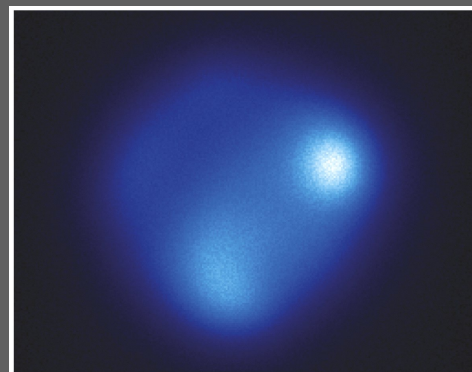
Telescope images (continued)

- The **blurring effects (seeing)** of atmospheric turbulence can be minimized by placing the telescope atop a tall mountain with very smooth air.
- They can be dramatically reduced by the use of **adaptive optics** and can be eliminated entirely by placing the telescope in orbit

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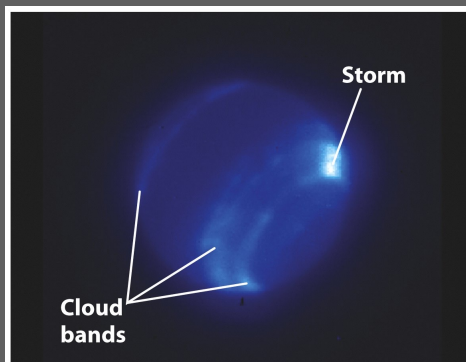


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(a) Neptune viewed without adaptive optics

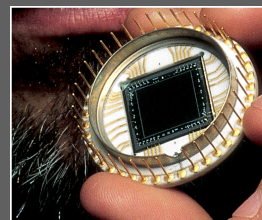
38



(b) Neptune viewed with adaptive optics

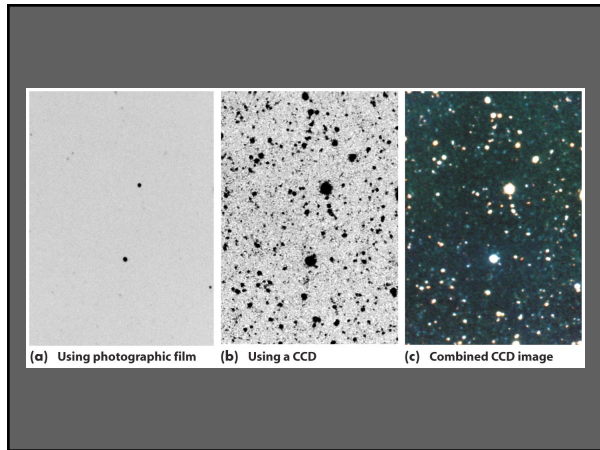
39

An electronic device is commonly used to record the image at a telescope's focus

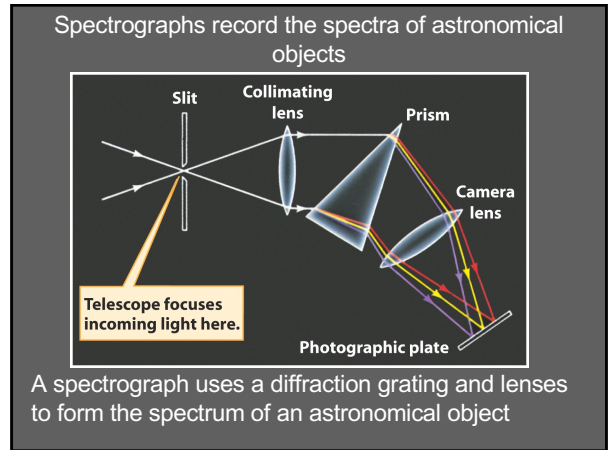


- Sensitive light detectors called charge coupled devices (CCDs) are often used at a telescope's focus to record faint images.

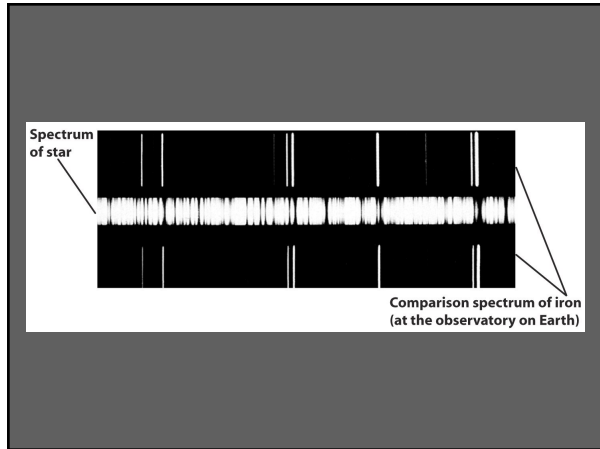
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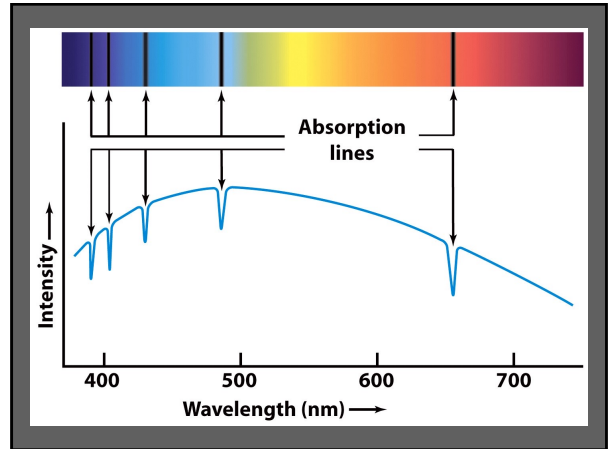
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43



44

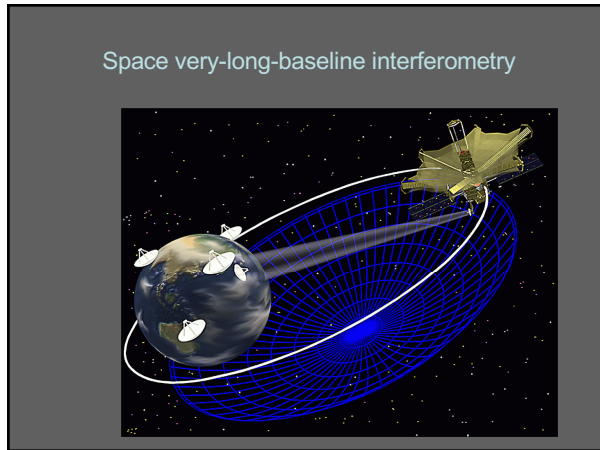
A radio telescope uses a large concave dish to reflect radio waves to a focus

- Radio telescopes use large reflecting antennas or dishes to focus radio waves
- Very large dishes provide reasonably sharp radio images

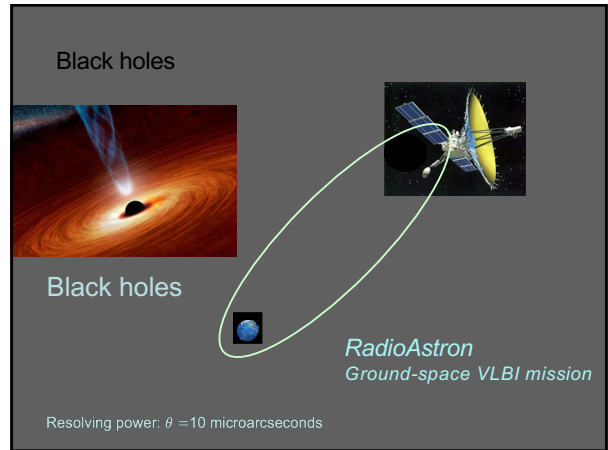
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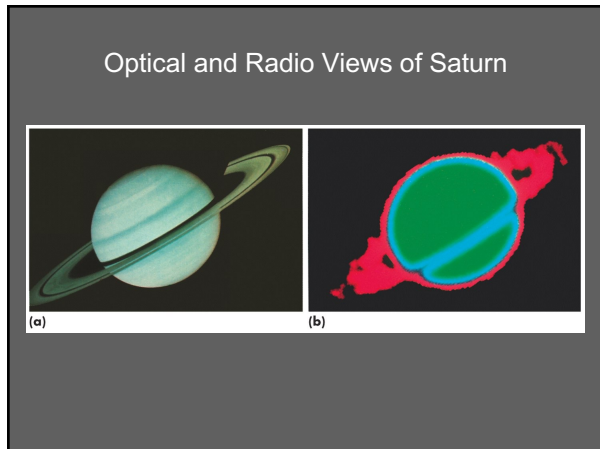
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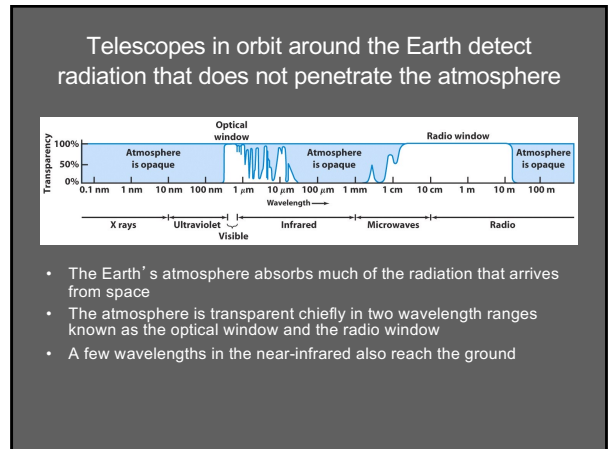
47



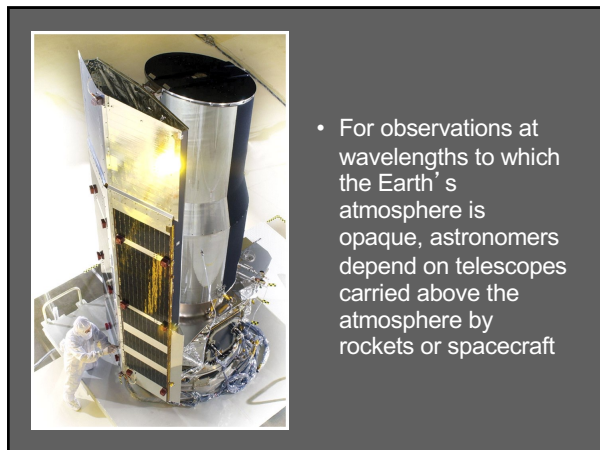
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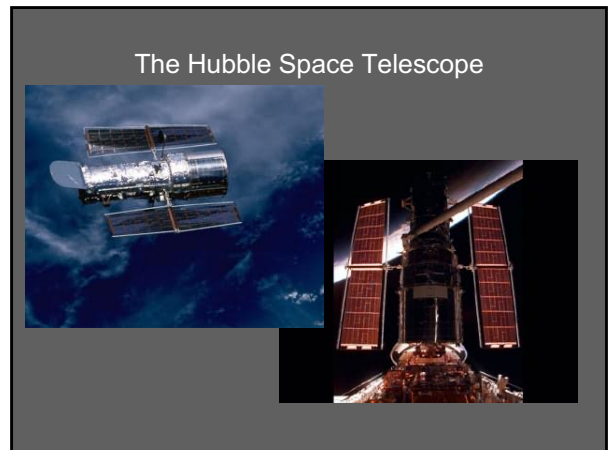
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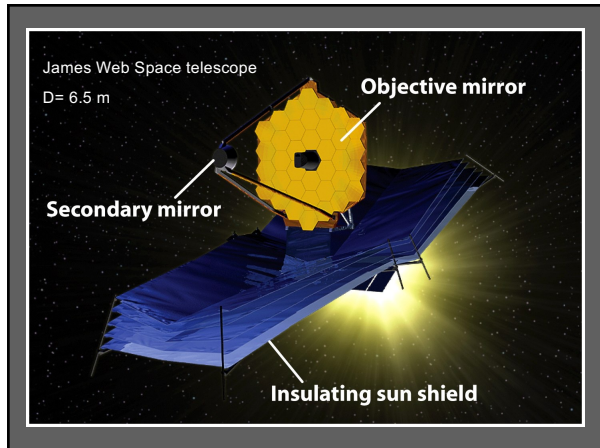
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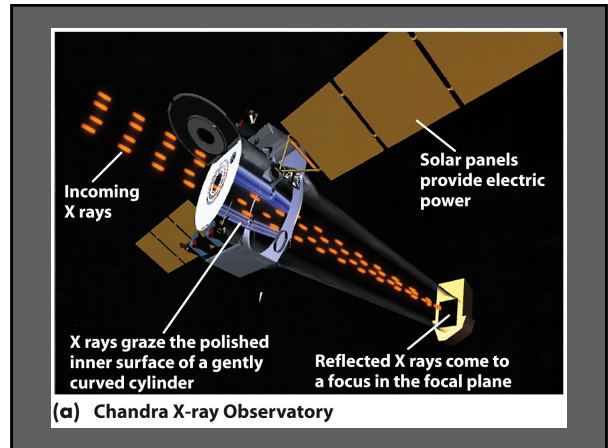
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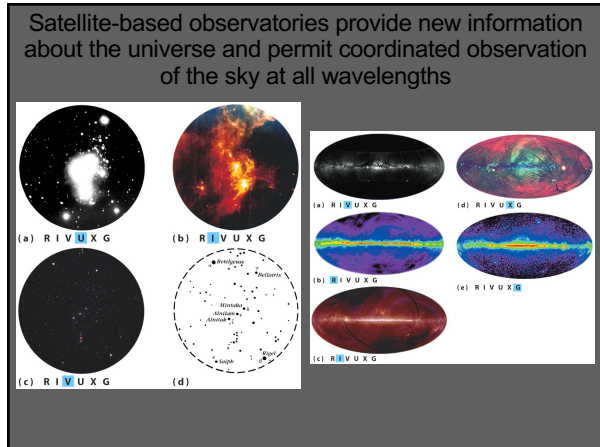
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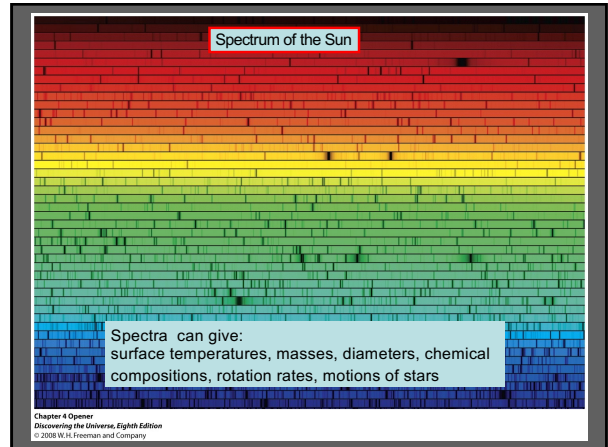
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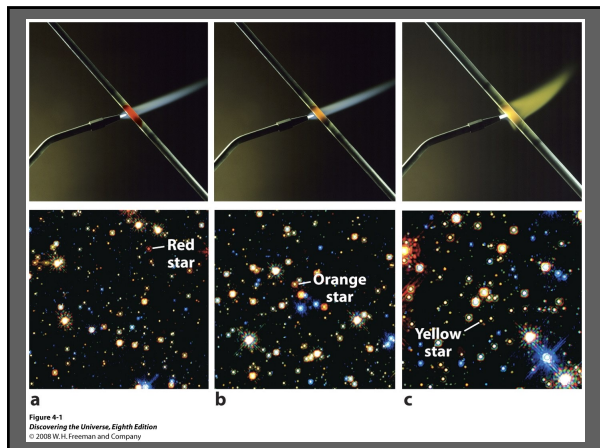
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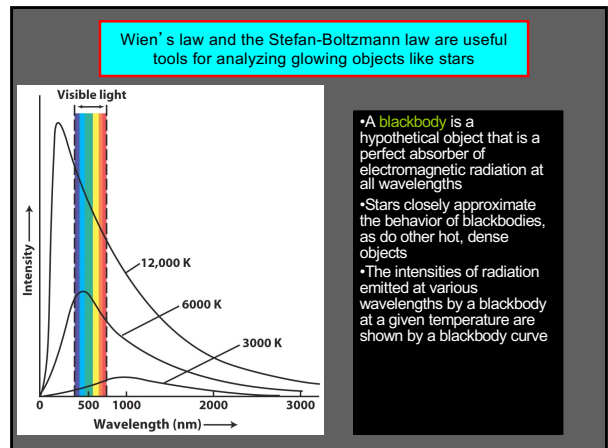
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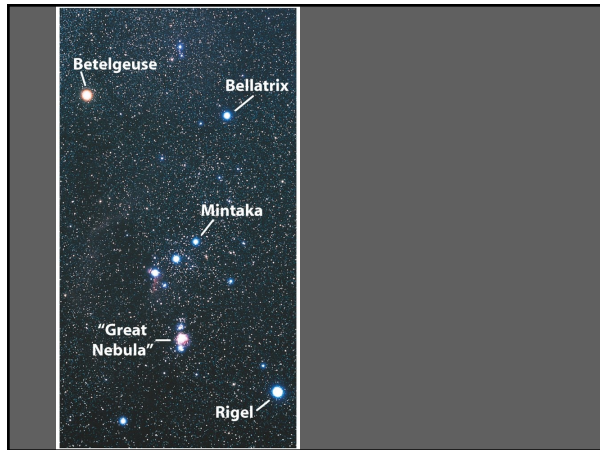
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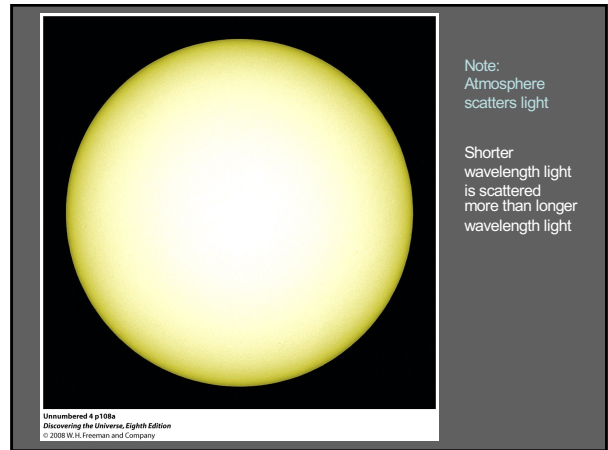
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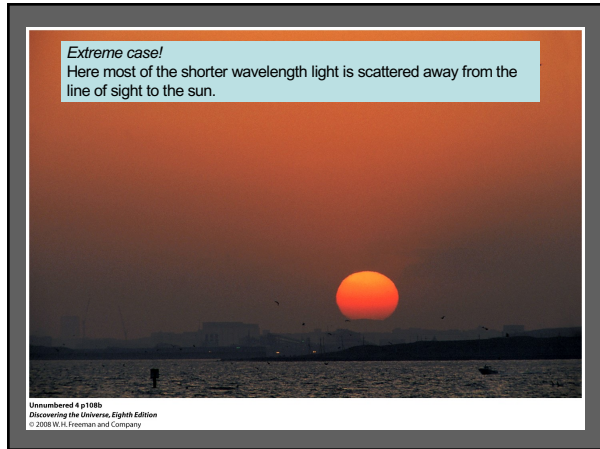
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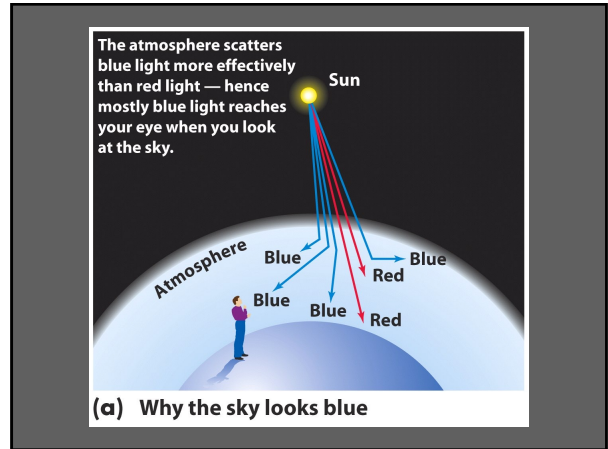
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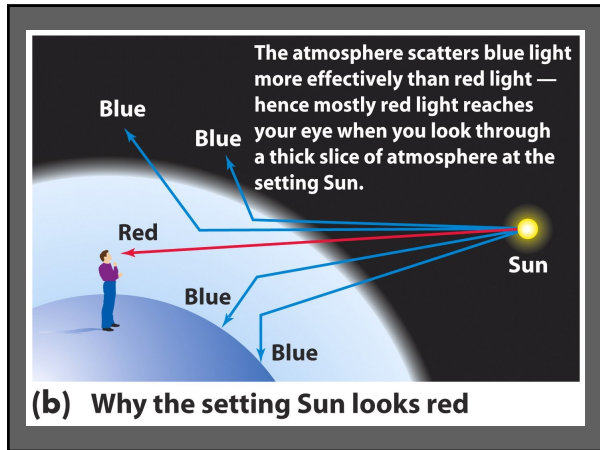
61



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63



64

Wien's Law for a Blackbody ...

$$\lambda_{\text{max}} = \frac{0.0029 \text{ K m}}{T}$$

λ_{max} = wavelength of maximum emission of the object (in meters)

T = temperature of the object (in kelvins)

• Example:
 • $\lambda_{\text{max}} = 500 \text{ nm}$, then
 • $T = 0.0029 \text{ K m} / 500 \text{ nm}$
 • $= 0.0029 / 5.0 \cdot 10^{-7} \text{ K}$
 • $= 5800 \text{ K}$

The higher the temperature of a blackbody, the shorter the wavelength of maximum emission (the wavelength at which the curve peaks).

The higher the temperature of a blackbody, the more light is emitted at all wavelengths.

Intensity

Wavelength (nm)

Visible light

12,000 K

6000 K

3000 K

66

Stefan-Boltzmann Law

The Stefan-Boltzmann law states that a blackbody radiates electromagnetic waves with a total energy flux F (watts per square metre) directly proportional to the fourth power of the Kelvin temperature T of the object:

$$F = \sigma T^4$$

F = energy flux in Joules per second per square meter of surface of object

$$\sigma = 5.670 \cdot 10^{-8} \text{ W m}^{-2} \text{ K}^{-4}$$

T = object's surface temperature in K

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Luminosity, an intrinsic quantity

Luminosity L (watts) is the total energy emitted by a star every second.

If we know how much energy is emitted every second from a 1m^2 patch on the star (from the Stefan-Boltzmann Law), then we can easily calculate the total energy emitted every second from the entire star's surface.

Multiplying the flux from the 1m^2 patch by the star's whole surface area:

$$L = 4\pi R^2 \times \sigma T^4$$

where R = radius of the star in m

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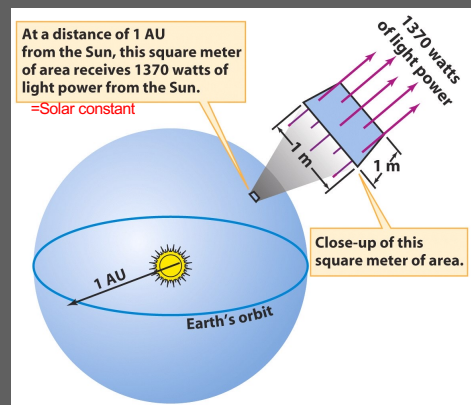
Example: What is the luminosity of the Sun?

$T = 5780 \text{ K}$ (Surface temperature of the Sun)

$R_{\text{sun}} = 696,000 \text{ km}$ (Radius of Sun)

$$\begin{aligned} L &= 4\pi R^2 \times \sigma T^4 \\ &= 4\pi \times (6.96 \times 10^8)^2 \times 5.670 \cdot 10^{-8} \times 5780^4 \\ L &= 3.85 \times 10^{26} \text{ W} \quad (\text{Luminosity of Sun}) \end{aligned}$$

69



70

Example: What is the power per square meter received from the Sun at Earth's distance?

$T = 5780 \text{ K}$

$R_{\text{sun}} = 696,000 \text{ km}$

$$\begin{aligned} L &= 4\pi R^2 \times \sigma T^4 \quad (\text{Luminosity}) \\ &= 4\pi \times (6.96 \times 10^8)^2 \times 5.670 \cdot 10^{-8} \times 5780^4 \\ &= 3.85 \times 10^{26} \text{ W} \end{aligned}$$

$$\begin{aligned} F_d &= L / (4\pi d^2) \quad (\text{Flux at distance } d \text{ from celestial object}) \\ &= 3.85 \times 10^{26} / (4\pi \times (1.5 \times 10^{11})^2) \\ &= 1360 \text{ W/m}^2 \end{aligned}$$

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Light has property of wave and particle

Energy of a photon: **•Example:**

$$E = hc/\lambda$$

$$= h\nu$$

$$h = 6.67 \cdot 10^{-34} \text{ Js}$$

(Planck's constant)

•Green light: $\lambda = 500 \text{ nm}$

$$\begin{aligned} \bullet E &= 6.67 \cdot 10^{-34} \cdot 3 \cdot 10^8 / (5 \cdot 10^{-7}) \\ \bullet &= 4.00 \cdot 10^{-19} \text{ J} \end{aligned}$$

72

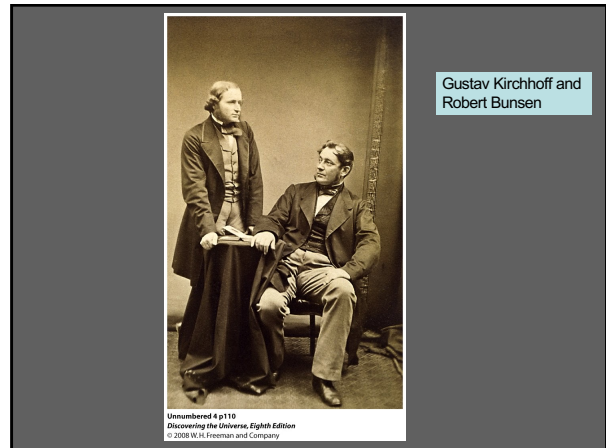
TABLE 4-1 Some Properties of Electromagnetic Radiation

	Wavelength (nm)	Photon energy (eV)*	Blackbody temperature (K)
Radio	$>10^7$	$<10^{-4}$	<0.03
Microwave**	10^5 to 4×10^5	10^{-4} to 3×10^{-3}	0.03 to 30
Infrared	4×10^3 to 7×10^2	3×10^{-3} to 2	30 to 4100
Visible	7×10^2 to 4×10^1	2 to 3	4100 to 7300
Ultraviolet	4×10^2 to 10^1	3 to 10^3	7300 to 3×10^4
X ray	10^3 to 10^{-2}	10^3 to 10^5	3×10^4 to 3×10^6
Gamma ray	$<10^{-2}$	$>10^5$	$>3 \times 10^6$

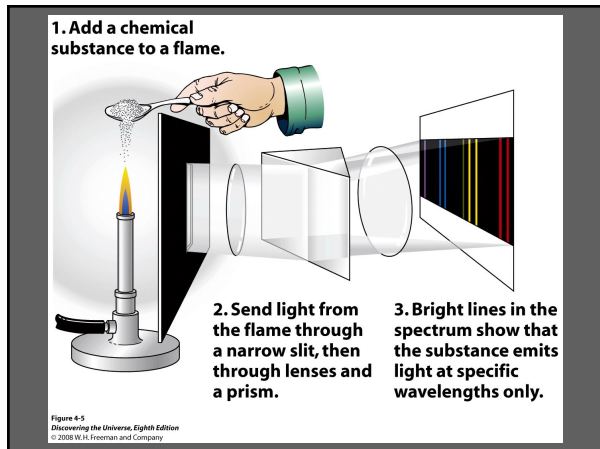
Note: $>$ means greater than; $<$ means less than.
 *1 eV = 1.6×10^{-19} J.
 **Microwaves, listed here separately, are often classified as radio waves or infrared radiation.

Table 4-1
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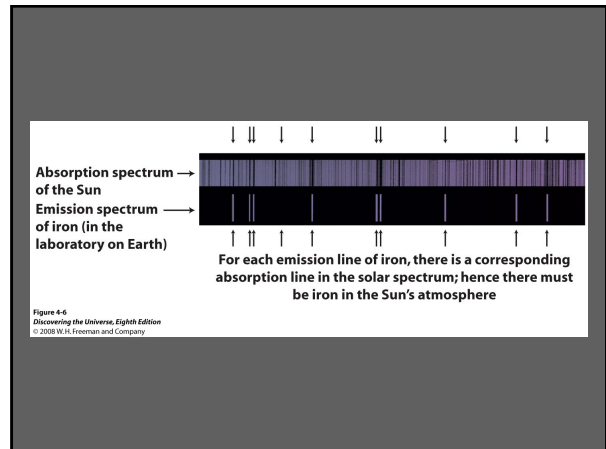
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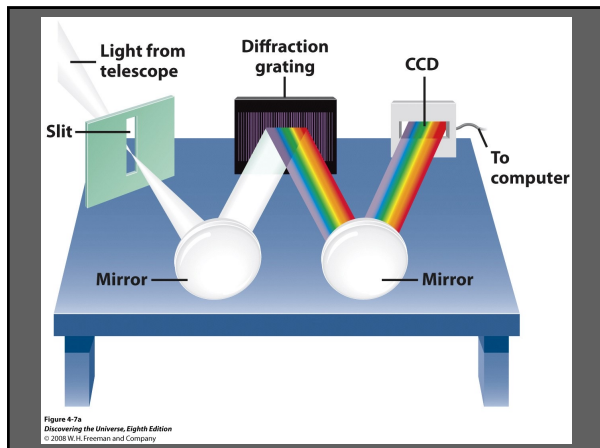
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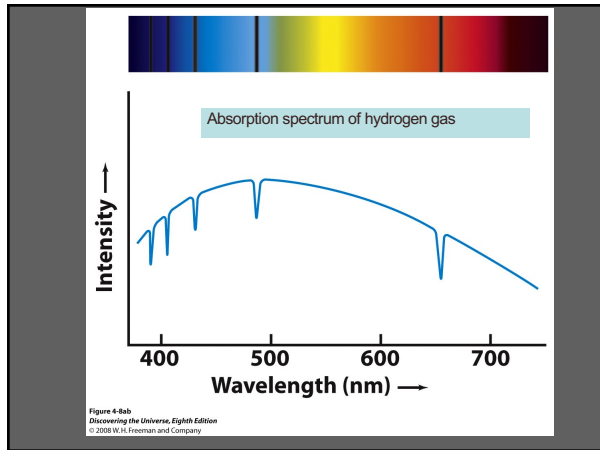
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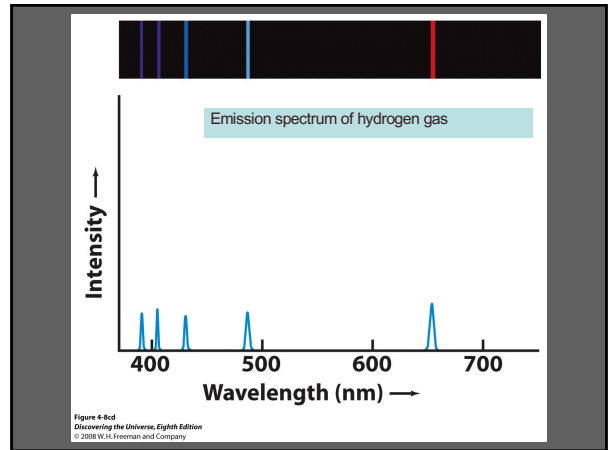
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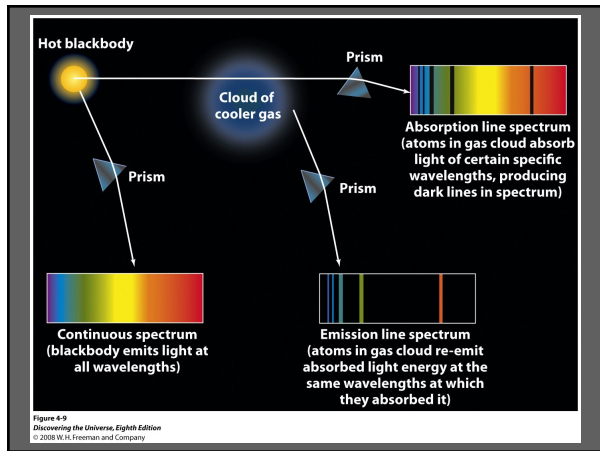
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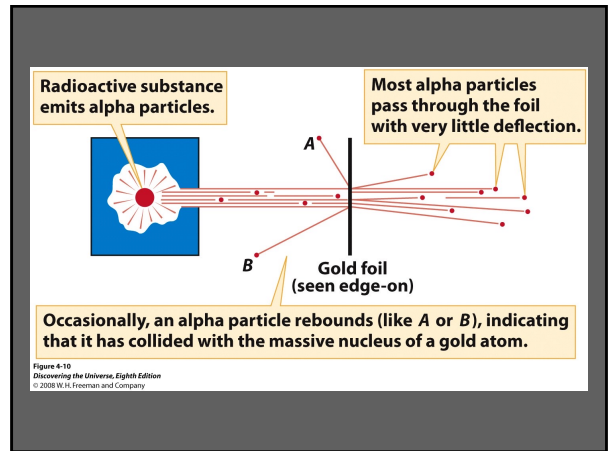
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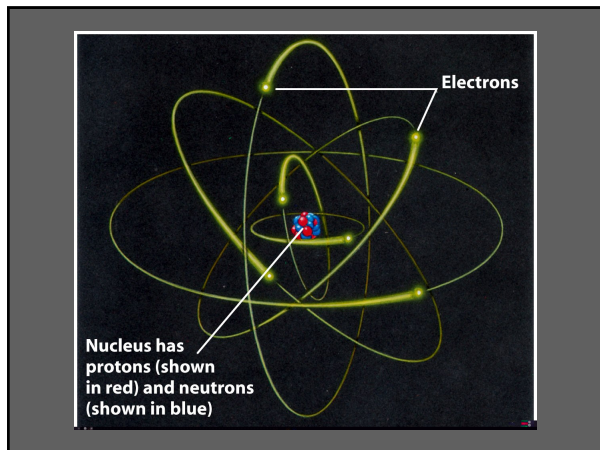
80



81



82



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Periodic Table of the Elements

Atomic number = number of protons
Different numbers of neutrons ==> isotopes
Unstable isotopes = radioactive isotopes

1	H	2	He																																
3	Li	4	Be	5	B	6	C	7	N	8	O	9	F	10	Ne																				
11	Na	12	Mg	13	Al	14	Si	15	P	16	S	17	Cl	18	Ar																				
19	K	20	Ca	21	Sc	22	Ti	23	V	24	Cr	25	Mn	26	Fe	27	Co	28	Ni	29	Cu	30	Zn	31	Ga	32	Ge	33	As	34	Se	35	Br	36	Kr
37	Rb	38	Sr	39	Y	40	Zr	41	Nb	42	Mo	43	Tc	44	Ru	45	Rh	46	Pd	47	Ag	48	Cd	49	In	50	Sn	51	Sb	52	Te	53	I	54	Xe
55	Cs	56	Ba	57	La	58	Ce	59	Pr	60	Nd	61	Pm	62	Sm	63	Eu	64	Gd	65	Tb	66	Dy	67	Ho	68	Er	69	Tm	70	Yb				
87	Fr	88	Ra	89	Ac	90	Th	91	Pa	92	U	93	Np	94	Pu	95	Am	96	Cm	97	Bk	98	Cf	99	Es	100	Fm	101	Md	102	No				

•The number of protons in an atom's nucleus is the **atomic number** for that particular element. It determines the element.
•The same element may have different numbers of neutrons in its nucleus.
•These slightly different kinds of elements are called **isotopes**.

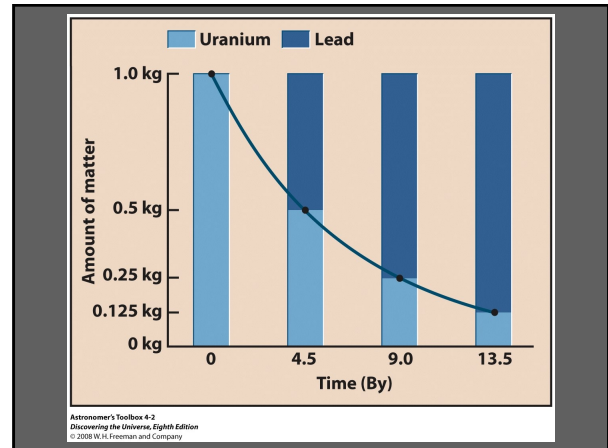
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TABLE 4-2 The Four Fundamental Forces of Nature

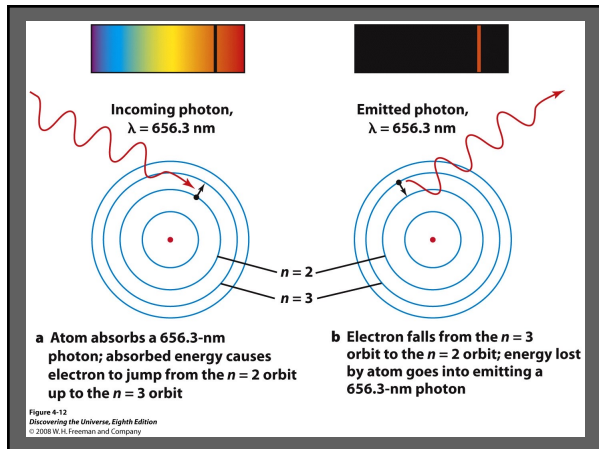
Name	Strength (compared to the strong force)	Range of effect (from each object)
Strong force	1	Inside atomic nuclei
Electromagnetic force	$1/137$	Throughout the universe
Weak force	10^{-6}	Inside atomic nuclei
Gravitational force	6×10^{-39}	Throughout the universe

Table 4-2
Discovering the Universe, Eighth Edition
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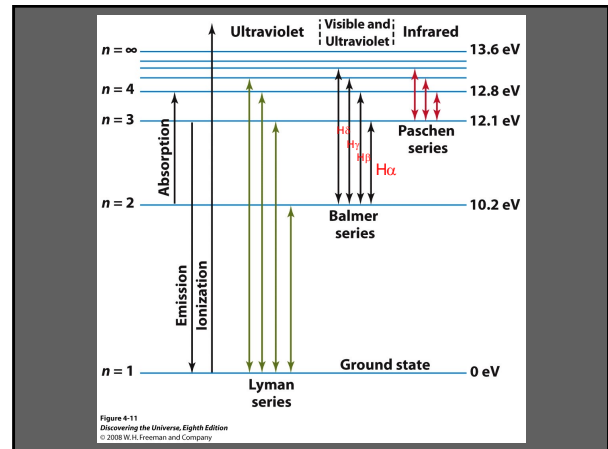
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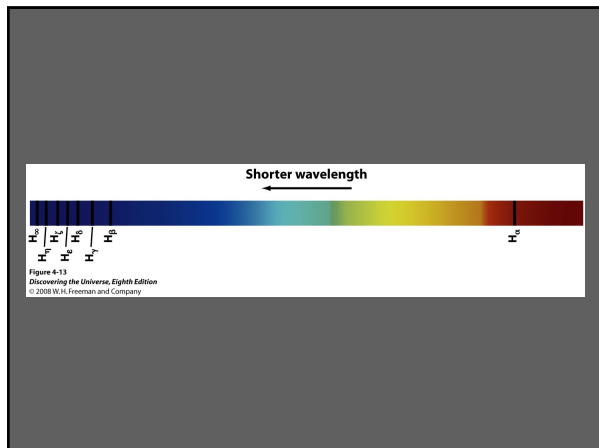
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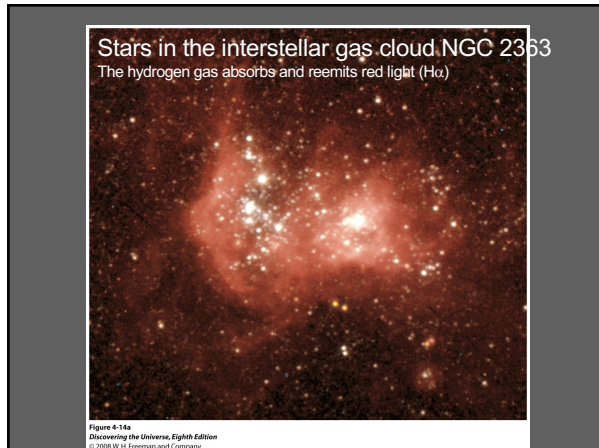
Spectral lines are produced when an electron jumps from one energy level to another within an atom

Classical Bohr model of an atom

- The nucleus of an atom is surrounded by electrons that occupy only certain orbits or energy levels
- When an electron jumps from one energy level to another, it emits or absorbs a photon of appropriate energy (and hence of a specific wavelength). $E = hc/\lambda$.
- The spectral lines of a particular element correspond to the various electron transitions between energy levels in atoms of that element.

Allowed Bohr orbits:
 $n = 1$
 $n = 2$
 $n = 3$
 $n = 4$
etc.

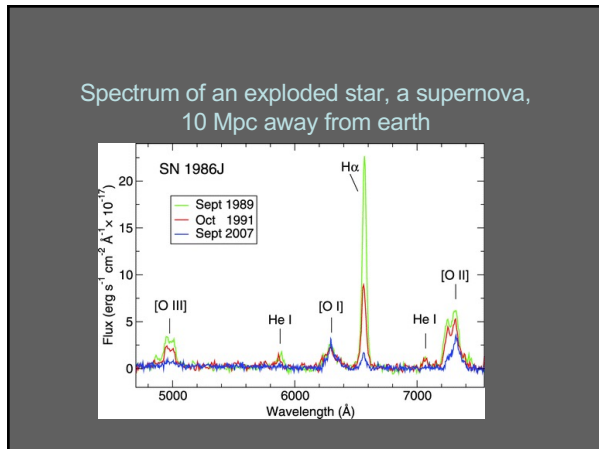
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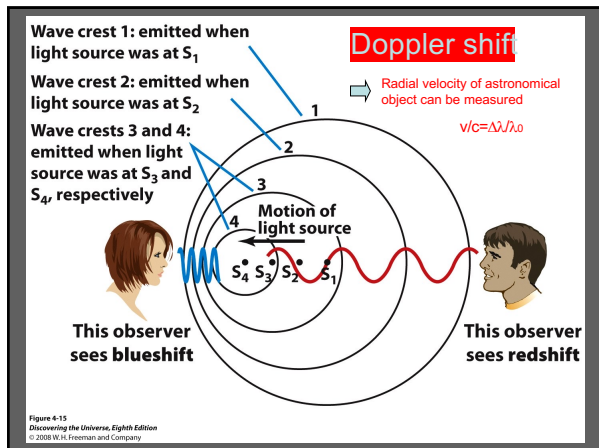
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• Christian Doppler 1803 - 1853

$$\frac{\Delta\lambda}{\lambda_0} = \frac{v}{c}$$

$\Delta\lambda$: wavelength shift
 λ_0 : wavelength if source is at rest
 v : radial velocity of source
 c : speed of light

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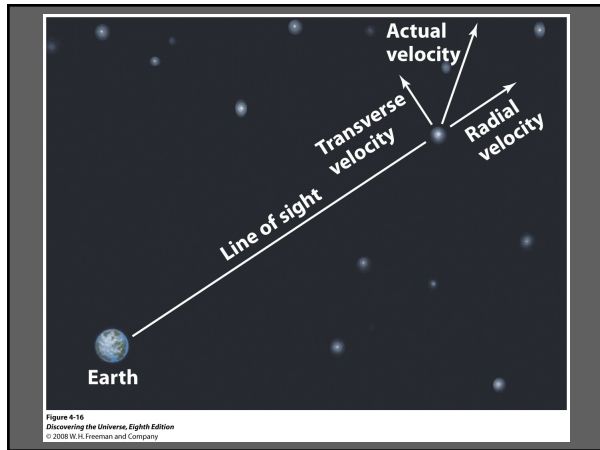


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Doppler Shifts

- **Red Shift:** The object is moving away from the observer (+ velocity). Wavelength increases.
- **Blue Shift:** The object is moving towards the observer (- velocity). Wavelength decreases.

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