Biophysics I (BPHS 4080)

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Website: http://www.yorku.ca/cberge/4080W2018.html

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Reference/Acknowledgement: - TF Weiss (Cellular Biophysics) - D Freeman

$$\frac{1}{2\pi a(r_o + r_i)} \frac{\partial^2 V_m}{\partial z^2} = C_m \frac{\partial V_m}{\partial t} + G_K(V_m, t) (V_m - V_K) + G_{Na}(V_m, t) (V_m - V_{Na}) + G_L(V_m - V_L)$$

$$G_K(V_m, t) = \overline{G}_K n^4(V_m, t)$$

$$G_{Na}(V_m, t) = \overline{G}_{Na} m^3(V_m, t) h(V_m, t)$$

$$n(V_m, t) + \tau_n(V_m) \frac{dn(V_m, t)}{dt} = n_\infty(V_m)$$

$$m(V_m, t) + \tau_m(V_m) \frac{dm(V_m, t)}{dt} = m_\infty(V_m)$$

$$h(V_m, t) + \tau_h(V_m) \frac{dh(V_m, t)}{dt} = h_\infty(V_m)$$

Question: So what do *m*, *h*, and *n* physically represent?

$$\tau_{x} \frac{dx}{dt} + x = x_{\infty} \qquad \frac{dx}{dt} = \alpha_{x}(1-x) - \beta_{x}x$$
$$x_{\infty} = \alpha_{x}/(\alpha_{x} + \beta_{x}) \text{ and } \tau_{x} = 1/(\alpha_{x} + \beta_{x})$$

$$\begin{aligned} \alpha_m &= \frac{-0.1(V_m + 35)}{e^{-0.1(V_m + 35)} - 1}, \\ \beta_m &= 4e^{-(V_m + 60)/18}, \\ \alpha_h &= 0.07e^{-0.05(V_m + 60)}, \\ \beta_h &= \frac{1}{1 + e^{-0.1(V_m + 30)}}, \\ \alpha_n &= \frac{-0.01(V_m + 50)}{e^{-0.1(V_m + 50)} - 1}, \\ \beta_n &= 0.125e^{-0.0125(V_m + 60)}, \end{aligned}$$



 \rightarrow Notion of an *ion channel*

<u>Question</u>: So what do *m*, *h*, and *n* physically represent?

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Intracellular







Figure 12.14: (Schematic; optical micrograph.) (a) A small patch of membrane containing only a single voltagegated sodium channel (or a few) is electrically isolated from the rest of the cell by a patch electrode. The current entering the cell through these channels is recorded by a monitor connected to the patch electrode. (b) Patch-clamp manipulation of a single, live photoreceptor cell from the retina of a salamander. The cell is secured by partially sucking it into a glass micropipette (bottom), while the patch-clamp electrode (upper left) is sealed against a small patch of the cell's plasma membrane. [Digital image kindly supplied by T. D. Lamb; see Lamb et al., 1986.]

Model: Voltage-Gated Two-State Molecular Gate

Note: The interplay between micro- & macro-scopic descriptions requires a transition into the domain of probability & expectation values



Figure 6.33 (mod)



→ Microscopic model (+ law of large numbers) gives rise to macroscopic behavior

<u>Question(s)</u>: How big must N be? How "local" does it need to be (i.e., as a channel density)?



<u>Model</u>: Voltage-Gated Two-State Molecular Gate (*Expected Values*)



Assume \mathcal{N} channels per unit area, of which n(t) are open.

n(*t*) is <u>average #</u> of open channels

$$n(t) = n_{\infty} + (n(0) - n_{\infty}) e^{-t/\tau_x}; \quad n_{\infty} = \frac{\alpha}{\alpha + \beta} \mathcal{N}, \quad \tau_x = \frac{1}{\alpha + \beta}$$

 $rac{dn(t)}{dt} = lpha(\mathcal{N}-n(t)) - eta n(t)$

First-order kinetics(!!)

Assume \mathcal{N} is large.

$$x(t) = ext{probability gate is open} pprox rac{n(t)}{\mathcal{N}}$$

 $x(t) = x_{\infty} + (x(0) - x_{\infty}) e^{-t/\tau_x}; \quad x_{\infty} = rac{lpha}{lpha + eta}, \quad au_x = rac{1}{lpha + eta}$

Model: Voltage-Gated Two-State Molecular Gate



The potential energy of an ion channel includes mechanical, chemical, and electrical contributions, each of which can be different in different conformations. Electrical potential energy depends on both the distribution of charge in the gate and on transmembrane potential. Therefore, E_B , E_O , and E_C depend on V_m .

First-order kinetics variables





$$E_B = \frac{1}{2}QV_B$$
; $E_C = \frac{1}{2}QV_m$; $E_O = -\frac{1}{2}QV_m$

$$\alpha = A \ e^{\frac{1}{2}Q(V_m - V_B)/kT}; \quad \beta = A \ e^{-\frac{1}{2}Q(V_m + V_B)/kT}$$

<u>Ex.</u>

$$x_{\infty} = \frac{\alpha}{\alpha + \beta} = \frac{1}{1 + \beta/\alpha} = \frac{1}{1 + e^{-QV_m/kT}}$$

$$\tau_x = \frac{1}{\alpha + \beta} = \frac{1}{A(e^{\frac{1}{2}Q(V_m - V_B)/kT} + e^{-\frac{1}{2}Q(V_m + V_B)/kT})}$$
$$= \frac{1}{Ae^{-\frac{1}{2}QV_B/kT}(e^{\frac{1}{2}QV_m/kT} + e^{-\frac{1}{2}QV_m/kT})}$$









Separating Out the Gating Current







(i.e., $Q_{on} = Q_{off}$ implies charge is conserved)

Saturation

(i.e., finite number of channels)

Question: If we know the single channel conductance, can we estimate the total # of contributing channels?

Hodgkin Huxley model



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 $\overline{G}_{Na} = 120, \ \overline{G}_{K} = 36, \ \text{and} \ G_{L} = 0.3 \ \text{mS/cm}^{2};$ $c_{Na}^{o} = 491, \ c_{Na}^{i} = 50, \ c_{K}^{o} = 20.11 \ c_{K}^{i} = 400 \ \text{mmol/L};$ $C_{m} = 1 \ \mu\text{F/cm}^{2}; \ V_{L} = -49 \ \text{mV}; \ \text{temperature is } 6.3^{\circ}\text{C}.$





Extracellular



Extracellular









Multistate Channel Models





Molecular Underpinnings







Figure 6.70



Y. Jiang, A. Lee, J. Chen, V. Ruta, M. Cadene, B. Chalt, and R. MacKinnon (2003), Nature 423:33-41.





Figure 12.16: (Schematic; sketch.) (a) Conceptual model of a voltage-gated ion channel. A spring normally holds a valve closed. An electric field pointing upward lifts the positively charged valve, letting water flow downward. (b) Sketch of the sodium channel. *Left:* In the resting state, positive charges in the channel protein's four "sensing" alpha helices are pulled downward, toward the negative cell interior. The sensing helices in turn pull the channel into its closed conformation. *Right:* Upon depolarization, the sensing helices are pulled upward. The channel now relaxes toward a new equilibrium, in which it spends most of its time in the open state. The lower blob depicts schematically the channel-inactivating segment. This attached object can move into the channel, blocking ion passage even though the channel itself is in its open conformation. [After Armstrong & Hille, 1998.]