

Biophysics I (BPHS 4080)

Instructors: Prof. Christopher Bergevin (cberge@yorku.ca)

Website: <http://www.yorku.ca/cberge/4080W2018.html>

Diffusion equation

1. Fick's First Law: $\phi = -D \frac{\partial c}{\partial x}$

+

2. Continuity Equation: $\frac{\partial \phi}{\partial x} = -\frac{\partial c}{\partial t}$

$$\frac{\partial c}{\partial t} = D \frac{\partial^2 c}{\partial x^2}$$

(Fick's Second Law)

Diffusion processes

i.e., solutions to $\phi = -D \frac{\partial c}{\partial x}$ $\frac{\partial c}{\partial t} = D \frac{\partial^2 c}{\partial x^2}$

1. Equilibrium: Zero flux and concentration is independent of time

$D \neq 0 \Rightarrow$ concentration is independent of space and time

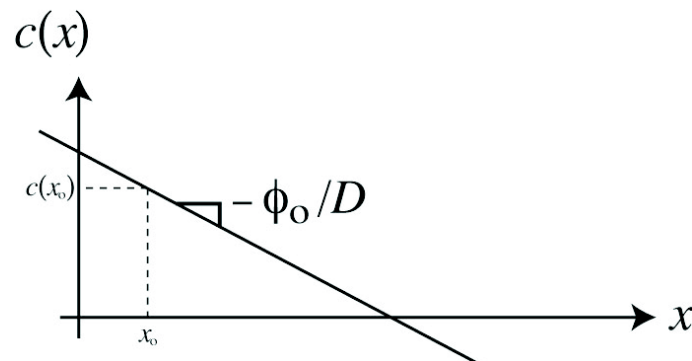
$D = 0 \Rightarrow$ non-diffusible solute is automatically at equilibrium

2. Steady-state: Flux can be non-zero, but flux and concentration are independent of time

$$\frac{\partial \phi}{\partial x} = 0 \quad \Rightarrow \quad \int \phi_o dx = \int -D dc \quad \Rightarrow \quad c(x) = c(x_o) - \frac{\phi_o}{D}(x - x_o)$$

[integrate Fick's 1st Law]

[x_o is a reference location where the concentration is known]



Diffusion processes

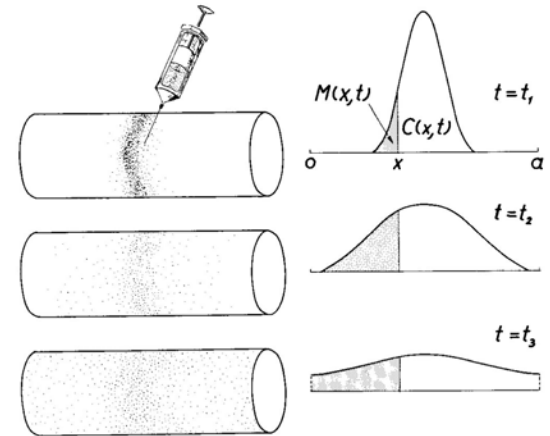
3. Impulse Response: Point-source of particles (n_o mol/cm²) at $t = 0$ and $x = 0$ [Dirac delta function $\delta(x)$]

given the initial/boundary conditions:

$$c(x, t) = n_o \delta(x) \quad \text{at } t = 0 \quad \text{where} \quad \int_{-\infty}^{\infty} \delta(x) dx = 1$$

need to solve:

$$\frac{\partial c}{\partial t} = D \frac{\partial^2 c}{\partial x^2}$$

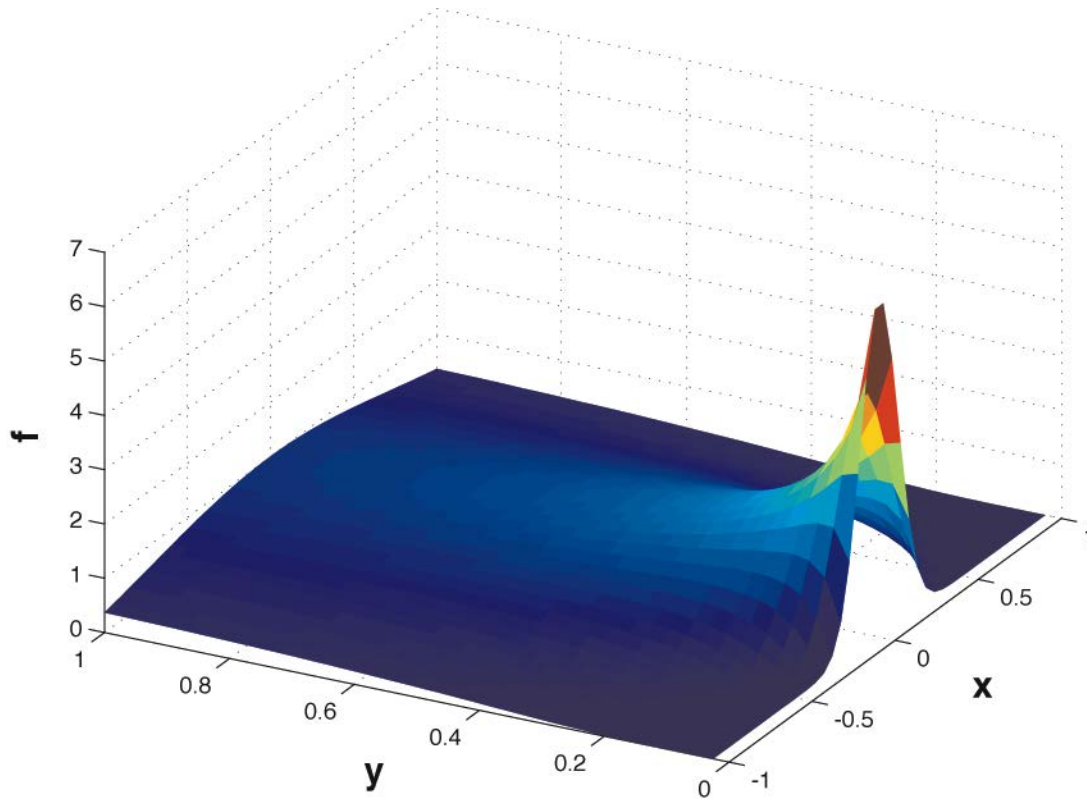


Batschelet Fig.12.5

[Aside: solution can be found by a # of different methods, one being by separation of variables and using a Fourier transform]

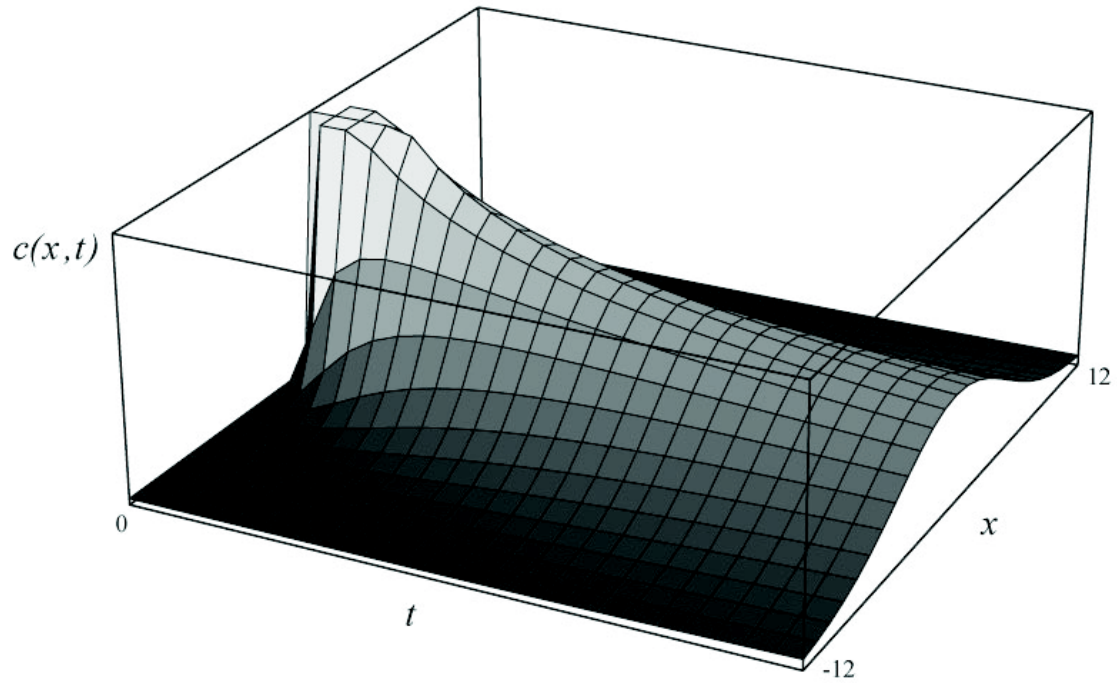
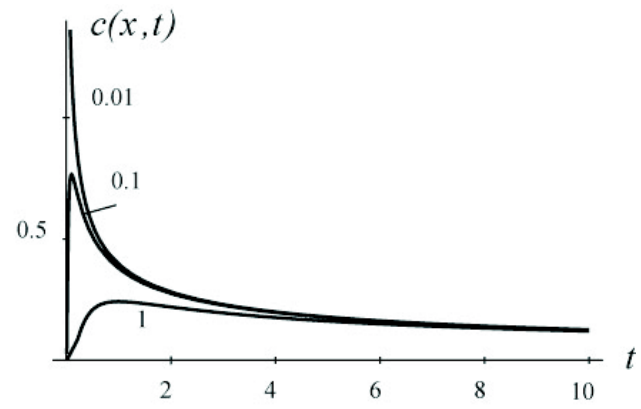
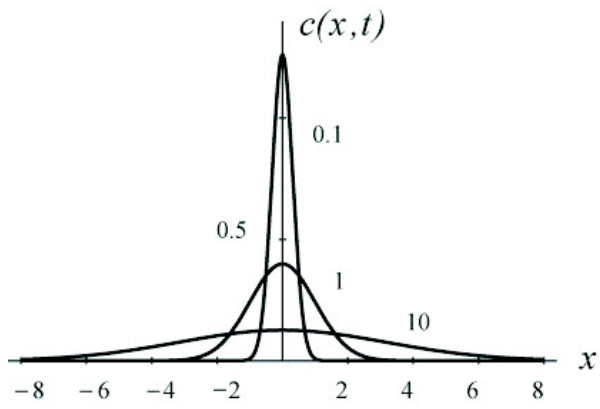
Solution
(for $t > 0$)

$$c(x, t) = \frac{n_o}{\sqrt{4\pi Dt}} e^{-x^2 / 4Dt}$$



$$f(x, y) = \frac{1}{\sqrt{y}} e^{-x^2/y}$$

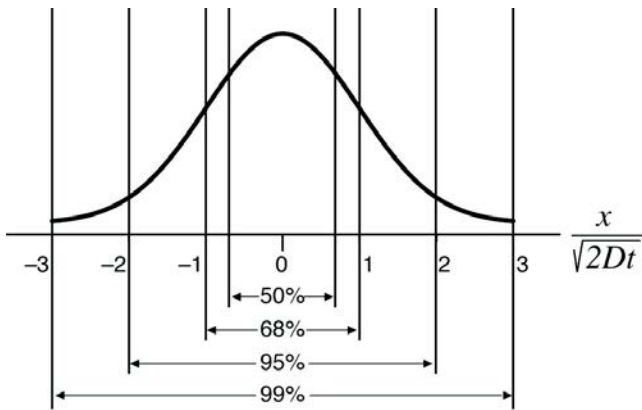
solution to
diffusion equation!



Importance of Scale

$$c(x, t) = \frac{n_o}{\sqrt{4\pi Dt}} e^{-x^2/4Dt}$$

Gaussian function with zero mean and standard deviation:
 $\sigma = \sqrt{2Dt}$



Question: How long does it take ($t_{1/2}$) for $\sim 1/2$ the solute to move at least the distance $x_{1/2}$?

$$\frac{x_{1/2}}{\sqrt{2Dt_{1/2}}} \approx \frac{2}{3} \implies t_{1/2} \approx \frac{x_{1/2}^2}{D}$$

For small solutes (e.g. K^+ at body temperature) $D \approx 10^{-5} \frac{\text{cm}^2}{\text{s}}$

	$x_{1/2}$	$t_{1/2}$
membrane sized	10 nm	$\frac{1}{10} \mu\text{sec}$
cell sized	10 μm	$\frac{1}{10} \text{sec}$
dime sized	10 mm	$10^5 \text{sec} \approx 1 \text{day}$

Exercise

At a junction between two neurons, called a synapse, there is a 20 nm cleft that separates the cell membranes. A chemical transmitter substance is released by one cell (the pre-synaptic cell), diffuses across the cleft, and arrives at the membrane of the other (post-synaptic) cell. Assume that the diffusion coefficient of the chemical transmitter substance is $D = 5 \times 10^{-6} \text{ cm}^2/\text{s}$.

→ Make a *rough estimate of the delay caused by diffusion of the transmitter substance across the cleft. What are the limitations of this estimate? Explain.*

Exercise

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Answer

Consider the time it takes for $\frac{1}{2}$ to cross the cleft, then we have approximately 1 μs ($1 \times 10^{-6} \text{ s}$). However, this calculation:

- Ignores the cleft geometry (e.g., not infinite baths)
- There is nothing special about $\frac{1}{2}$ the solute here (perhaps only a few molecules are needed, or perhaps a lot are)

Exercise

To wiggle your big toe, neural messages travel along a single neuron that stretches from the base of your spine to your toe. Assume that the membrane of this neuron can be represented as a uniform cylindrical shell that encloses the intracellular environment, which is represented as a simple saline solution. The diameter of the shell is $10\ \mu\text{m}$ and the length is $1\ \text{m}$. Assume that 10^{-15} moles of dye are injected into the neuron at time $t = 0$ and at a point located in the center of the neuron, which we will refer to as the point $z = 0$. Assume that the dye diffuses across the radial dimension so quickly that the concentration of dye $c(z,t)$ depends only on the longitudinal direction z and time t . Assume that the diffusivity of the dye in the intracellular saline is $D = 10^{-7}\ \text{cm}^2/\text{s}$ and that the membrane is impermeant to the dye.

- Determine the amount of time t_1 required for 5% the injected dye to diffuse to points outside the region $-1\ \text{mm} < z < 1\ \text{mm}$.
- Determine the amount of time t_2 required for half the injected dye to diffuse to points outside the region $-1\ \text{mm} < z < 1\ \text{mm}$. Determine the ratio of t_2 to t_1 . Briefly explain the physical significance of this result.
- Determine the amount of time t_3 required for 5% the injected dye to diffuse to points outside the region $-10\ \text{mm} < z < 10\ \text{mm}$. Determine the ratio of t_3 to t_1 . Briefly explain the physical significance of this result.

Answers

→ Determine the amount of time t_1 required for 5% the injected dye to diffuse to points outside the region $-1 \text{ mm} < z < 1 \text{ mm}$.

3.5 hours

→ Determine the amount of time t_2 required for half the injected dye to diffuse to points outside the region $-1 \text{ mm} < z < 1 \text{ mm}$. Determine the ratio of t_2 to t_1 . Briefly explain the physical significance of this result.

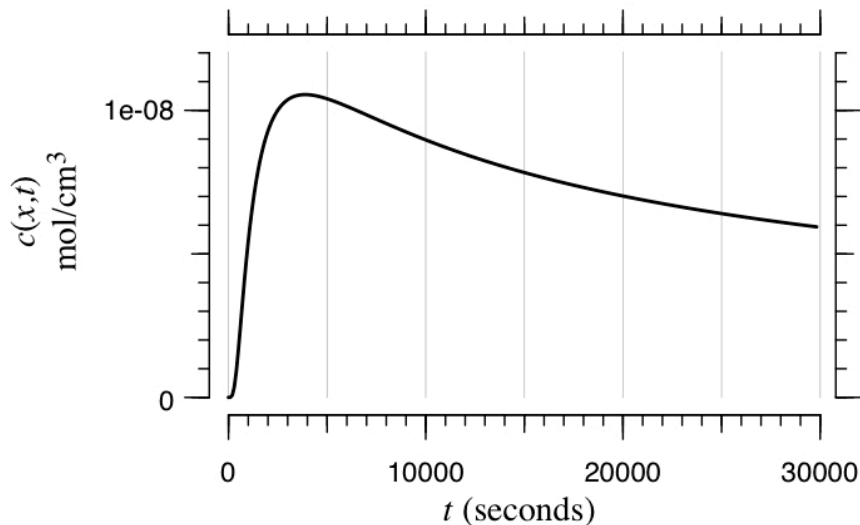
1.3 days

→ Determine the amount of time t_3 required for 5% the injected dye to diffuse to points outside the region $-10 \text{ mm} < z < 10 \text{ mm}$. Determine the ratio of t_3 to t_1 . Briefly explain the physical significance of this result.

14.5 days

Exercise

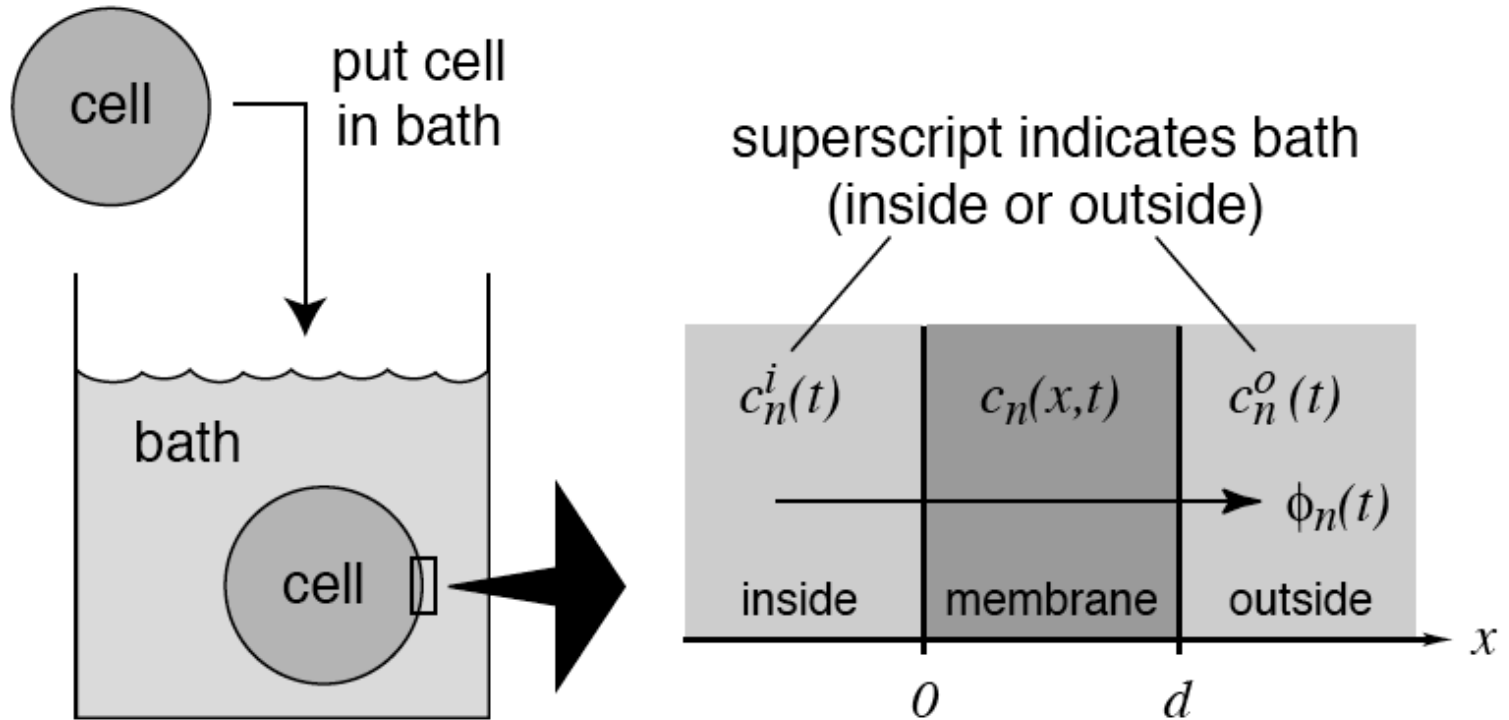
To wiggle your big toe, neural messages travel along a single neuron that stretches from the base of your spine to your toe. Assume that the membrane of this neuron can be represented as a uniform cylindrical shell that encloses the intracellular environment, which is represented as a simple saline solution. The diameter of the shell is $10\ \mu\text{m}$ and the length is $1\ \text{m}$. Assume that 10^{-15} moles of dye are injected into the neuron at time $t = 0$ and at a point located in the center of the neuron, which we will refer to as the point $z = 0$. Assume that the dye diffuses across the radial dimension so quickly that the concentration of dye $c(z,t)$ depends only on the longitudinal direction z and time t . Assume that the diffusivity of the dye in the intracellular saline is $D = 10^{-7}\ \text{cm}^2/\text{s}$ and that the membrane is impermeant to the dye.



The following plot shows the concentration of dye as a function of time for a particular point at $z_0 > 0$.

→ Determine z_0 .

Membrane Diffusion: Two-Compartment Geometry



reference direction for flux is outward

Diffusion Through Cell Membranes: History 101

Diffusion through Cell Membranes

Charles Ernest Overton (late 1800s): first systematic studies

- qualitative:
 - put cell in bath with solute
 - wait, rinse, squeeze
 - analyze to see how much got in (+ = some; +++ = a lot)
- 100's of solutes, dozens of cell types
- surprising results: previously cell membranes had been thought to be impermeant to essentially everything but water

Overton's Rules:

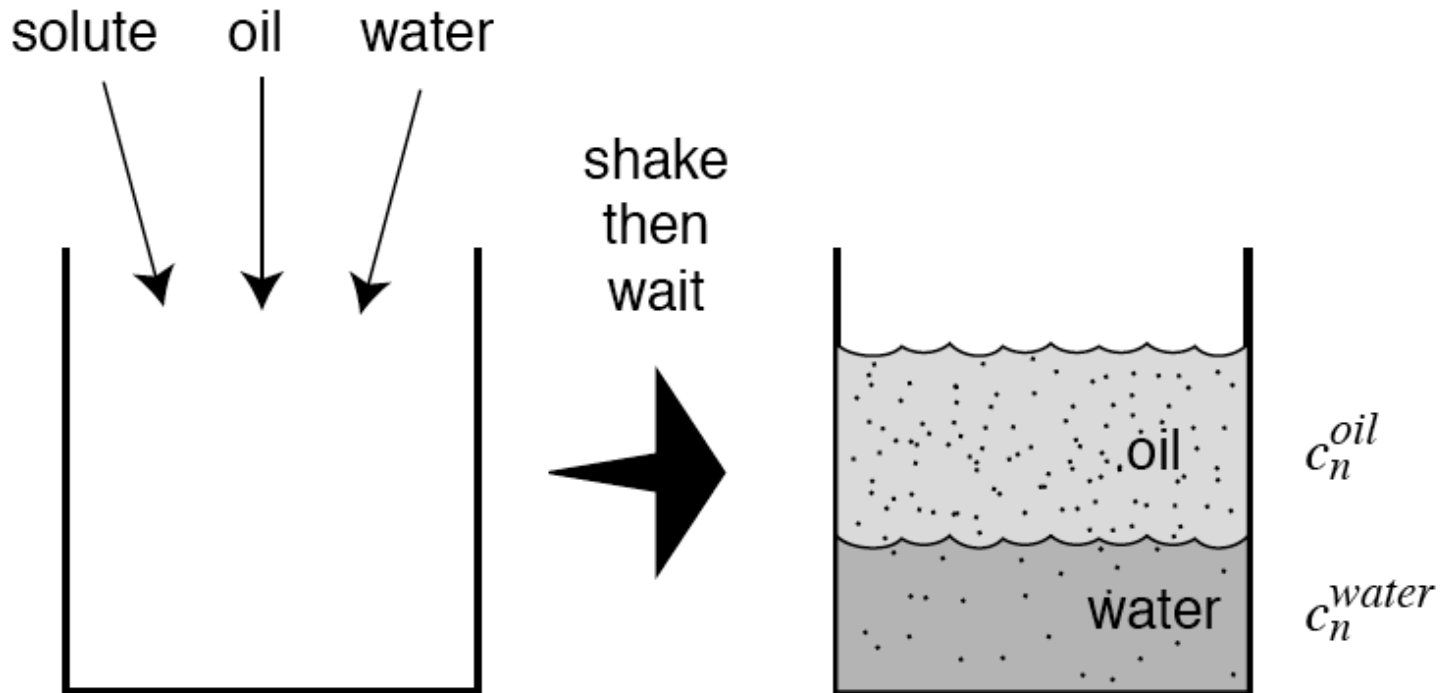
- cell membranes are semi-permeable
- relative permeabilities of plant and animals cells are similar
- permeabilities correlate with solubility of solute in organic solvents
 - membrane is lipid (specifically cholesterol and phospholipids)
- certain cells concentrate some solutes → active transport
- potency of anesthetics correlated with lipid solubility
 - Meyer-Overton theory of narcosis
- muscles don't contract in sodium-free media

Diffusion through Cell Membranes

Paul Runar Collander (1920-1950): first quantitative studies

- large cells (cylindrical algae cells, 1 mm diameter, 1 cm long)
- bathe cell in solute for time t_1 , squeeze out cytoplasm, analyze
- repeat with new cell and new time t_2
- plot intracellular quantity versus time
- fit with exponential function of time (two-compartment theory)
- infer permeability from time constant

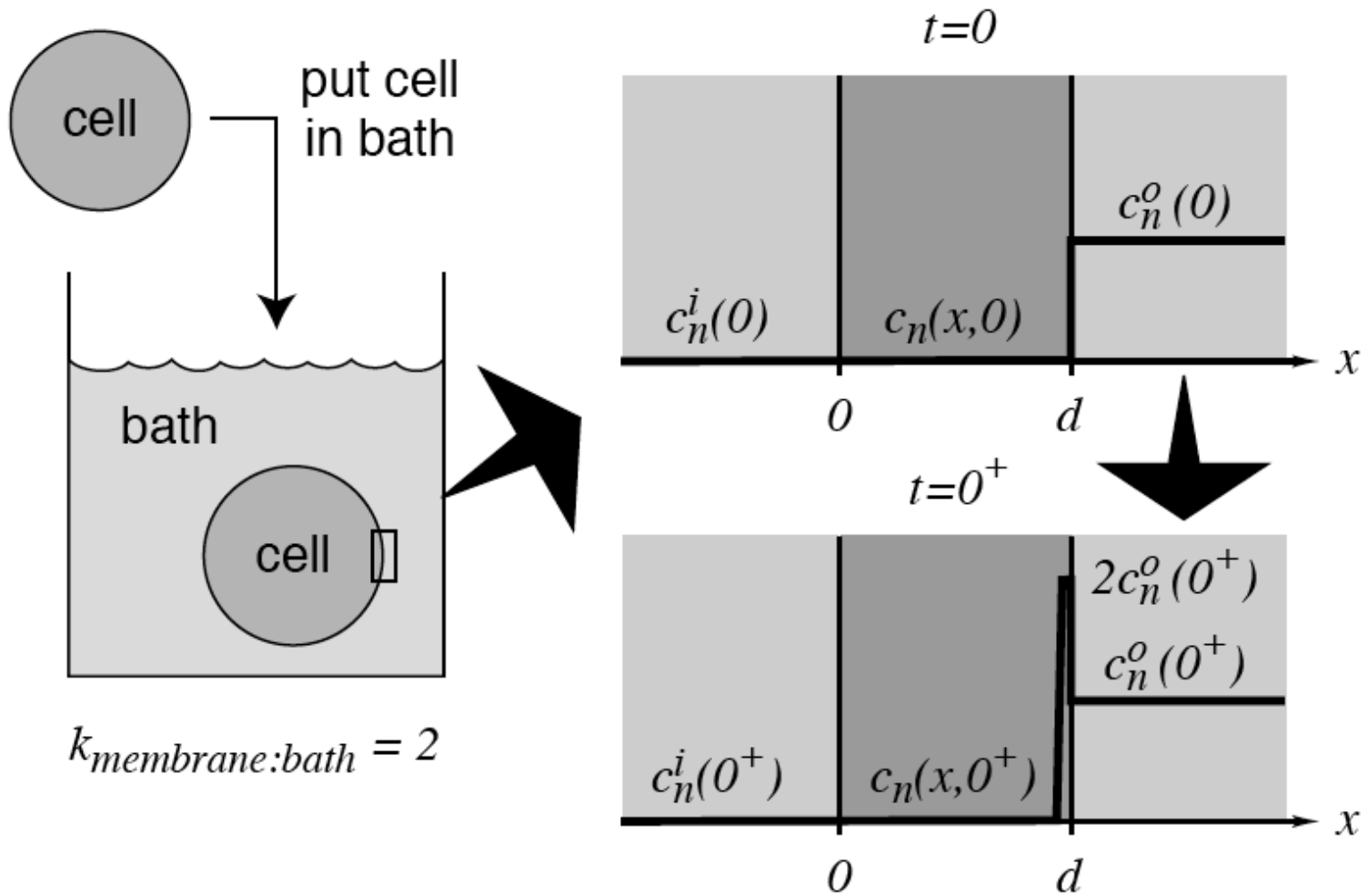
Step 1: Dissolve



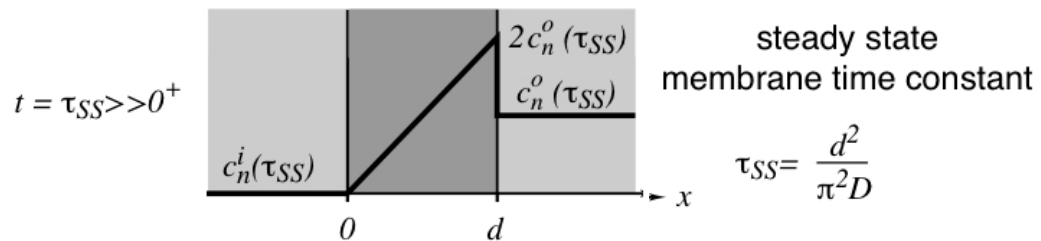
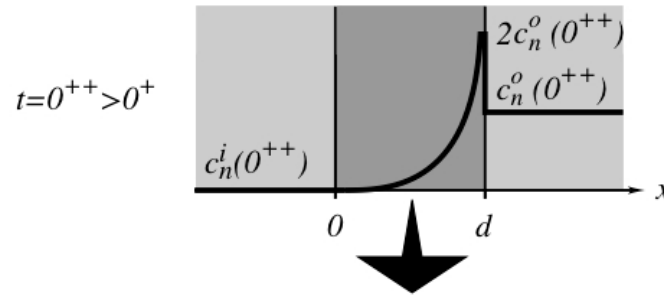
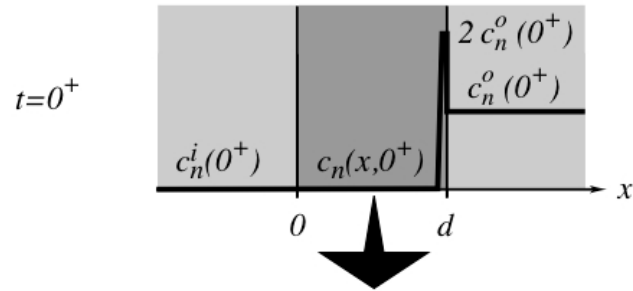
Equilibrium characterized by relative solubilities
of solute n in oil and water

$$\text{partition coefficient } k_{oil:water} = \frac{c_n^{oil}}{c_n^{water}}$$

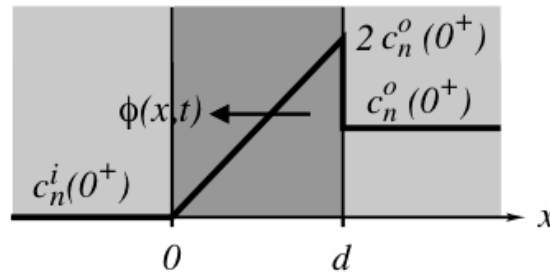
Assume Dissolving is fast relative to diffusing



Step 2: Solute diffuses through membrane



Step 3: Solute enters the cell



$$c_n(x,t) = c_n(0,t) + \frac{x}{d}(c_n(d,t) - c_n(0,t))$$

$$= k_n c_n^i(t) + \frac{k_n x}{d}(c_n^o(t) - c_n^i(t))$$

$$k_n = k_{\text{membrane:bath}}$$

$$= k_{\text{membrane:cytoplasm}}$$

Fick's law: $\phi_n(t) = -D_n \frac{\partial c_n(x,t)}{\partial x}$

$$= -D_n \frac{c_n(d,t) - c_n(0,t)}{d}$$

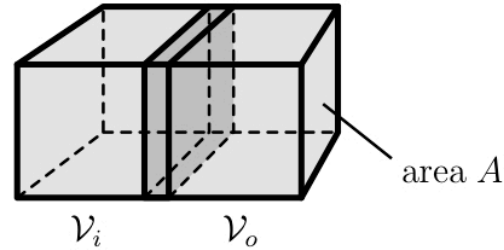
$$= \frac{D_n k_n}{d} (c_n^i(t) - c_n^o(t))$$

$$\phi_n(t) = P_n (c_n^i(t) - c_n^o(t)) ; P_n = \frac{D_n k_n}{d}$$

Fick's law for membranes

P_n = permeability of membrane to solute n

Step 4: Concentration in cell changes: two-compartment diffusion



Assume

- \mathcal{V}_i and \mathcal{V}_o constant
- well-stirred baths: $c_n^i(t)$, $c_n^o(t)$
- solute is conserved and membrane is thin: $c_n^i(t)\mathcal{V}_i + c_n^o(t)\mathcal{V}_o = N_n$
- membrane always in steady state: $\phi_n(t) = P_n(c_n^i(t) - c_n^o(t))$

By continuity,

$$A\phi_n(t) = -\frac{d}{dt}(c_n^i(t)\mathcal{V}_i) = \frac{d}{dt}(c_n^o(t)\mathcal{V}_o)$$

$$\frac{d}{dt}c_n^i(t) = -\frac{AP_n}{\mathcal{V}_i}(c_n^i(t) - c_n^o(t)) = -\frac{AP_n}{\mathcal{V}_i} \left(c_n^i(t) - \frac{1}{\mathcal{V}_o}N_n + c_n^i(t)\frac{\mathcal{V}_i}{\mathcal{V}_o} \right)$$

$$\frac{d}{dt}c_n^i(t) + AP_n\left(\frac{1}{\mathcal{V}_i} + \frac{1}{\mathcal{V}_o}\right)c_n^i(t) = \frac{AP_nN_n}{\mathcal{V}_i\mathcal{V}_o}$$

First-order linear differential equation with constant coefficients, therefore

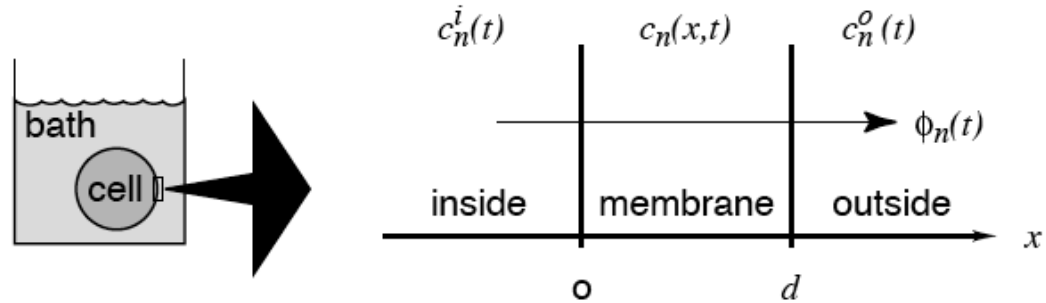
$$c_n^i(t) = c_n^i(\infty) + [c_n^i(0) - c_n^i(\infty)]e^{-t/\tau_{EQ}}$$

$$c_n^i(\infty) = \frac{N_n}{\mathcal{V}_i + \mathcal{V}_o}$$

$$\tau_{EQ} = \frac{1}{AP_n\left(\frac{1}{\mathcal{V}_i} + \frac{1}{\mathcal{V}_o}\right)}$$

Membrane Diffusion: Summary

Membrane diffusion



Dissolve and diffuse model

- solute outside cell dissolves into cell membrane
- solute diffuses through membrane
- solute dissolves into cytoplasm

Membrane time constant $t_{SS} = \frac{d^2}{\pi^2 D_n}$

Fick's law for membranes: $\phi_n(t) = P_n (c_n^i(t) - c_n^o(t))$; $P_n = \frac{D_n k_n}{d}$

Two-compartment diffusion

Cell time constant $t_{EQ} = \frac{1}{AP_n \left(\frac{1}{V_o} + \frac{1}{V_i} \right)}$

Dynamics of Membrane Diffusion

- Numerical solution to eqns.
- Arbitrary initial condition (top)
- Fast dynamics (middle)
- Steady-state set up (middle)
- Eventually, the two compartments change (bottom)

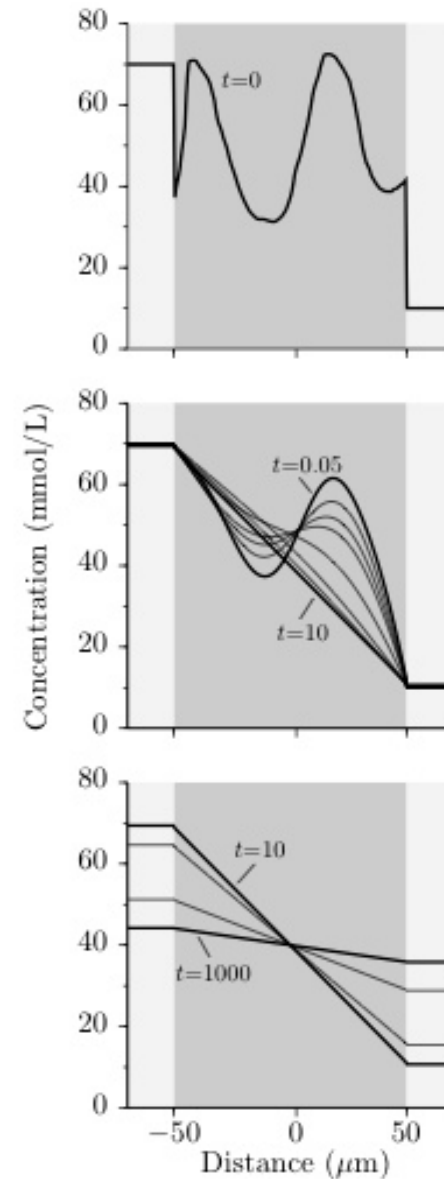
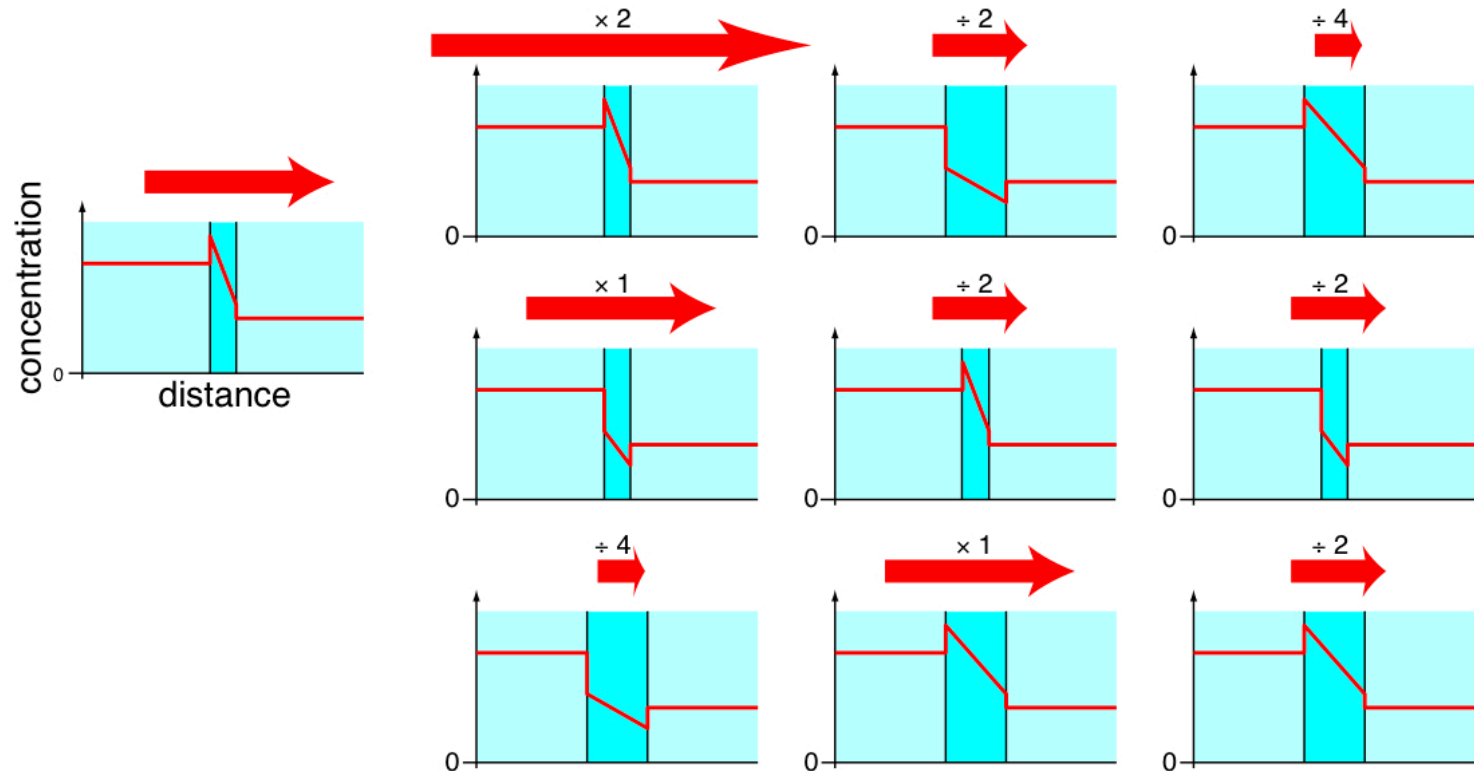


Figure 3.30

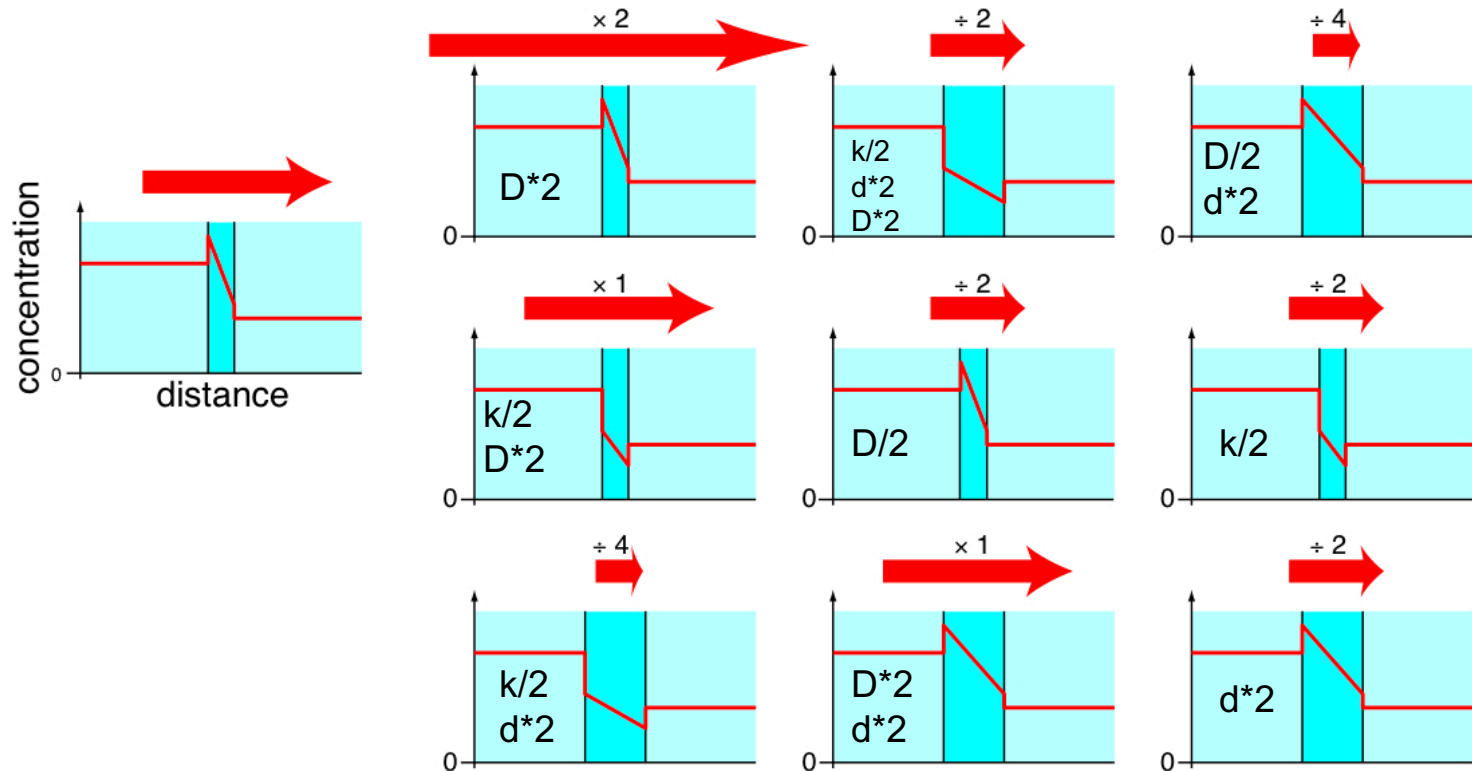
Effect of changing parameters on flux: What is being changed?



$$\phi_n(t) = P_n (c_n^i(t) - c_n^o(t)) ; P_n = \frac{D_n k_n}{d}$$

Fick's law for membranes

P_n = permeability of membrane to solute n



$$\phi_n(t) = P_n (c_n^i(t) - c_n^o(t)) ; P_n = \frac{D_n k_n}{d}$$

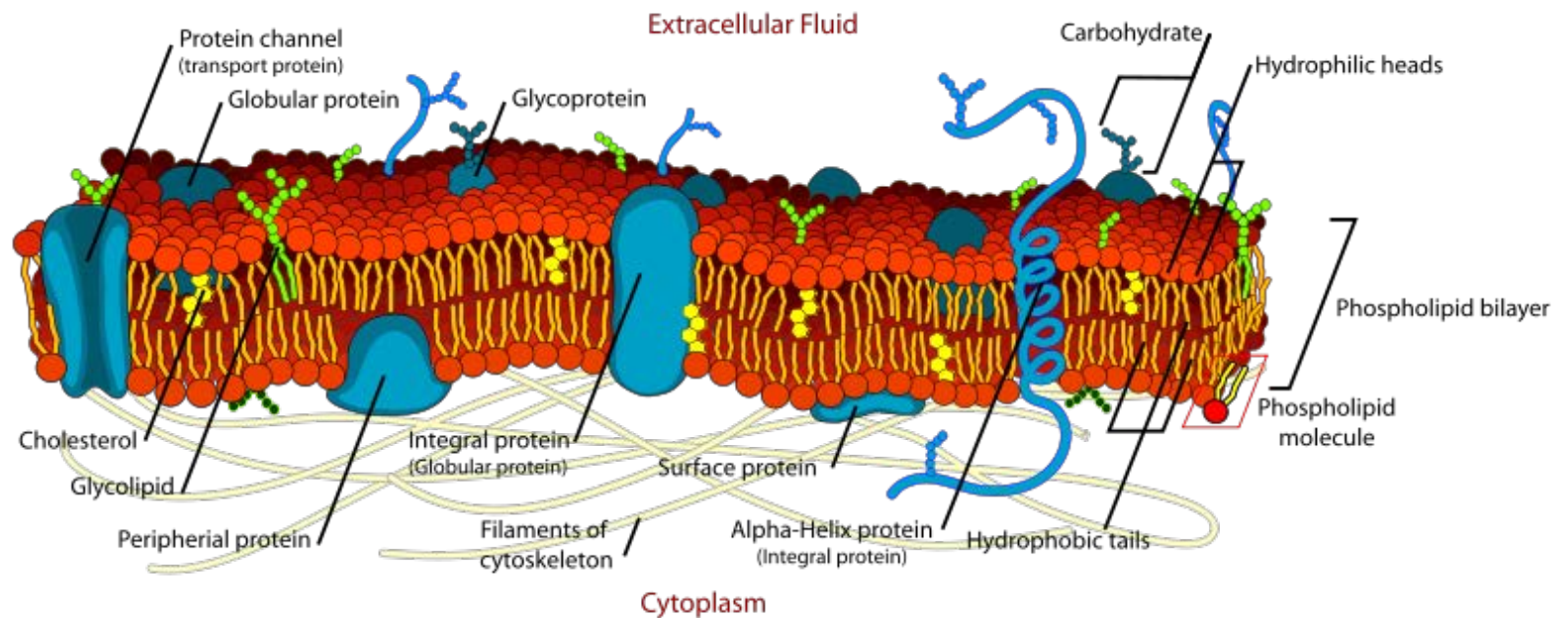
Fick's law for membranes

P_n = permeability of membrane to solute n

Question(s)

→ What are cell membranes made of?

→ How does one go about determining such?



→ It is only relatively recently we had a picture such as this!!