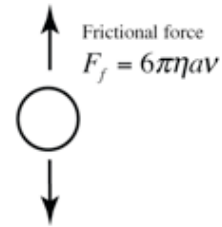


QUESTION TWO: Purcell seems unperturbed by the low efficiency of bacterial motility. Explain why so that even a non-physicist (like Dr. Lew) will understand. What percentage of the basal metabolic rate of a human would be required for propulsion if we had flagella and existed at a Reynolds number similar to that of a bacterium? Show your work.

The force that has to be overcome to maintain a constant velocity is a function of the viscosity of the solution and size of the object, Stokes Relation:

This drag is relatively small (compared to the higher drag that occurs at Reynolds numbers greater than 1, where turbulent drag starts to be a factor). And, since diffusion supplies more than sufficient energy to the bacterium, motility is not so metabolically expensive that the bacterium will starve.



To compare to humans, scale the energy requirement of 0.5 W per kilogram: about 35 W for a 70 kilogram human. A Watt is equal to a joule per second. There are $(60 \cdot 60 \cdot 24)$ [86400 sec/day] • 35 joules per second equals 3024000 joules per day. A kcal is equal to 4187 joules, so, about 725 kcal per day, compared to a basal metabolism of about 2000 kcal per day. It is significant for a human.

QUESTION THREE: At low Reynolds number, a jet propulsion mechanism won't work. Why? Propose an alternative mechanism —excluding flagella-based or pili twitching.

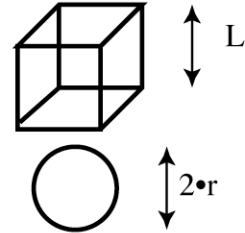
Because you have to 'pull' solution into the jet engine to eject it, it's reversible movement at low Reynolds number. The closest analogy in Purcell's paper is the scallop. Alternatives? This is a direct copy and paste from Purcell, who explores a number of mechanisms.

(End)

Symbol	Value	Units	Comments
GAS CONSTANT			
R	8.314	J mol ⁻¹ K ⁻¹	R is the Boltzmann constant times Avogadro's Number (6.023•10 ²³)
	1.987	cal mol ⁻¹ K ⁻¹	
	8.314	m ⁻³ Pa mol ⁻¹ K ⁻¹	
RT	2.437 • 10 ³	J mol ⁻¹	At 20 °C (293 °K)
	5.833 • 10 ²	cal mol ⁻¹	At 20 °C (293 °K)
	2.437	liter MPa mol ⁻¹	At 20 °C (293 °K)
RT/F	25.3	mV	At 20 °C (293 °K)
2.303 • RT	5.612	kJ mol ⁻¹	At 20 °C (293 °K)
	1.342	kcal mol ⁻¹	At 20 °C (293 °K)
k _B	1.381 • 10 ⁻²³	J K ⁻¹	Boltzmann constant
FARADAY CONSTANT			
F	9.649 • 10 ⁴	coulombs mol ⁻¹	F is the electric charge times Avogadro's Number
	9.649 • 10 ⁴	J mol ⁻¹ V ⁻¹	
	23.06	kcal mol ⁻¹ V ⁻¹	
CONVERSIONS			
kcal	4.187	kJ (kiloJoules)	Joules is an energy unit (equal to 1 Newton•meter)
erg (g•cm ² /s ²)	1 • 10 ⁻⁷	J (Joules)	(see above)
Watt	1	J sec ⁻¹	
Volt	1	J coulomb ⁻¹	
Amperes	1	coulomb sec ⁻¹	
Pascal (Pa)	1	Newton meter ⁻²	Pascal is a pressure unit (equal to 10 ⁻⁵ bars)
Radians	Radians•(180°/π)	degrees	Conversion of radians to degrees
PHYSICAL PROPERTIES			
η _w	1.004 • 10 ⁻³	Pa sec	viscosity of water at 20 °C
v _w	1.004 • 10 ⁻⁶	m ² sec ⁻¹	kinematic viscosity of water at 20 °C (viscosity/density)
V _w	1.805 • 10 ⁻⁵	m ³ mol ⁻¹	partial molal volume of water at 20 °C

Source: mostly Nobel, Park S (1991) Physicochemical and Environmental Physiology

A cube has a surface area of $6 \cdot L^2$. Its volume is L^3 . As long as the shape is constant, the ratio of surface area to volume will always be $(6 \cdot L^2) / L^3$, or $6/L$.



For a sphere, the surface area is $4 \cdot \pi \cdot r^2$, and the volume is $\pi \cdot r^3$; the corresponding ratio of surface area to volume is $4/r$.

density (water = 1 gm cm^{-3}) velocity (cm sec^{-1})

Reynolds number $R_e = \frac{\rho \cdot v \cdot l}{\eta}$ length (cm)

$$m \left(-\frac{dv}{dt} \right) = 6 \cdot \pi \cdot \eta \cdot r \cdot v$$

$$v(t) = v_0 e^{\left(-\frac{6 \cdot \pi \cdot \eta \cdot r \cdot t}{m} \right)}$$

$$F = Av + B\omega$$

$$N = Cv + D\omega$$

That is, both velocity and rotation contribute to both the force and torque.



↑ Frictional force

$$F_f = 6\pi\eta av$$

$$J_x = -D \frac{\partial c}{\partial x} + v_x \cdot c$$

units: $\text{moles cm}^{-2} \text{ sec}^{-1}$ $(\text{cm sec}^{-1})(\text{moles cm}^{-3})$
 $(\text{cm}^2 \text{ sec}^{-1})(\text{moles cm}^{-4})$

○ Where the frictional and gravitational forces are balanced, the velocity reaches a steady state.

↓ Gravitational pull

$$F_g = \frac{4}{3} \pi a^3 \Delta \rho g$$

Sherwood number $S_h = \frac{v \cdot l}{D}$

velocity (cm sec^{-1})
length (cm)
Diffusion ($\text{cm}^2 \text{ sec}^{-1}$)