

SC/BIOL 4151 Membrane Transport: Second Hour Test

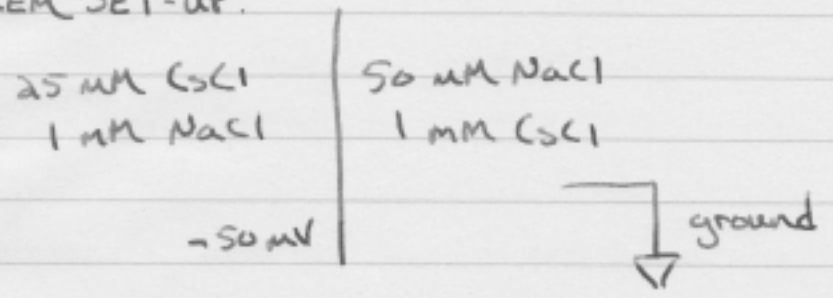
To assess the selectivity of the gramicidin channel to different cations, you add 25 mM CsCl plus 1 mM NaCl to one side (*cis*) of the bilayer chamber setup, and 50 mM NaCl plus 1 mM CsCl to the other side (*trans*). The *trans* chamber containing NaCl is the 'ground', with a voltage of zero relative to the *cis* chamber containing CsCl. You measure a zero current potential (where flux is zero) of -50 mV.

Determine the ratio of $P_{\text{Cs}}/P_{\text{Na}}$. And, interpret your result based upon the diffusion coefficients for Cs^+ ($2.06 \cdot 10^{-5}$ cm/sec) and Na^+ ($1.33 \cdot 10^{-5}$ cm/sec)

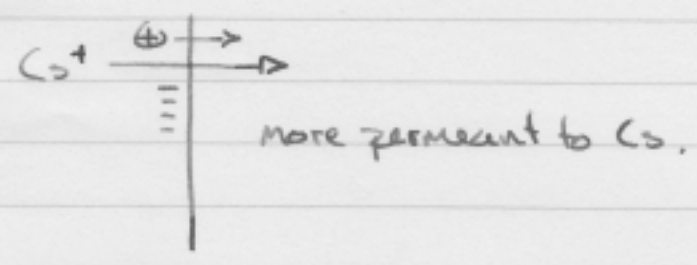
Please assume (correctly) that when the *cis* chamber contains 100 mM KCl and the *trans* chamber contains 10 mM KCl, the zero current potential is about -55 mV.

POINTS

PROBLEM SET-UP.



(4)



(2)

EQUATION SET-UP

$$\psi = \frac{RT}{F} \ln \frac{P_{Na} [Na^{trans}] + P_{Cs} [Cs^{trans}]}{P_{Na} [Na^{cis}] + P_{Cs} [Cs^{cis}]} \quad (4)$$

$P_{Cl} [Cl]$ is not included. KCl experiment demonstrates $P_{Cl} = 0$ (1)

ALGEBRA.

$$\frac{[Na^{trans}] + \frac{P_{Cs}}{P_{Na}} [Cs^{trans}]}{[Na^{cis}] + \frac{P_{Cs}}{P_{Na}} [Cs^{cis}]} \quad (1)$$

RESULT: $\frac{P_{Cs}}{P_{Na}} \sim 20$ (1)

INTERPRETATION. Ratio significantly higher than evidence of selectivity by the channel, separate from properties of the ions alone.

$\frac{D_{Cs}}{D_{Na}}$
 (1)
 (14)

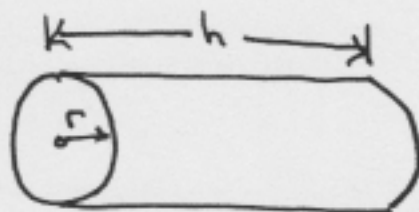
EQUATIONS

Sphere Area: $4\pi r^2$

Sphere Volume: $\frac{4}{3}\pi r^3$

Cylinder Area: $2\pi r h$

Cylinder Volume: $\pi r^2 h$



GOLDMAN EQUATION

$$J = -P \frac{zF\psi}{RT} \left[\frac{c^o - c^i e^{\left(\frac{zF\psi}{RT}\right)}}{1 - e^{\left(\frac{zF\psi}{RT}\right)}} \right]$$

zero flux $\psi = \frac{RT}{zF} \ln \left[\frac{c^o}{c^i} \right]$

zero potential $J = -P(c^o - c^i)$

net charge $Q = C \Delta E$

(coulombs) (volt)

capacitance per unit area ($\sim 1 \mu\text{F}/\text{cm}^2$)

for a spherical cell: $C = 4\pi r^2 C'$

$$Q = \frac{4}{3}\pi r^3 C F$$

concentration

Goldman-Hodgkin-Katz equation (assuming only Na^+ , K^+ , and Cl^- are present):

$$\psi = \frac{RT}{F} \ln \left[\frac{P_{\text{Na}} C_{\text{Na}}^o + P_{\text{K}} C_{\text{K}}^o + P_{\text{Cl}} C_{\text{Cl}}^i}{P_{\text{Na}} C_{\text{Na}}^i + P_{\text{K}} C_{\text{K}}^i + P_{\text{Cl}} C_{\text{Cl}}^o} \right]$$

Goldman-Hodgkins-Katz equation (assuming a single cation (M^+) and anion (A^-) are present):

$$\psi = \frac{RT}{F} \ln \left[\frac{[\text{M}^+]_i + \frac{P_{\text{A}^-}}{P_{\text{M}^+}} [\text{A}^-]_o}{[\text{M}^+]_o + \frac{P_{\text{A}^-}}{P_{\text{M}^+}} [\text{A}^-]_i} \right]$$